

# 2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester

**Other names:** Methacrylic acid, 2-(diethylamino)ethyl ester  
«beta»-(Diethylamino)ethyl methacrylate  
«beta»-(N,N-Diethylamino)ethyl methacrylate  
Daktose B  
Diethylaminoethyl methacrylate  
Ethanol, 2-(diethylamino)-, methacrylate (ester)  
N,N-Diethylaminoethyl methacrylate  
2-(N,N-Diethylamino)ethyl methacrylate  
2-Diethylaminoethyl methacrylate  
2-Diethylaminoethylester kyseliny methakrylove  
NSC 14490

**Inchi:** InChI=1S/C10H19NO2/c1-5-11(6-2)7-8-13-10(12)9(3)4/h3,5-8H2,1-2,4H3

**InchiKey:** SJIXRGNQPBQWMK-UHFFFAOYSA-N

**Formula:** C10H19NO2

**SMILES:** C=C(C)C(=O)OCCN(CC)CC

**Mol. weight [g/mol]:** 185.26

**CAS:** 105-16-8

## Physical Properties

Property code	Value	Unit	Source
gf	-10.53	kJ/mol	Joback Method
hf	-311.36	kJ/mol	Joback Method
hfus	24.87	kJ/mol	Joback Method
hvap	48.46	kJ/mol	Joback Method
log10ws	-1.29		Crippen Method
logp	1.447		Crippen Method
mcvol	164.880	ml/mol	McGowan Method
pc	2295.91	kPa	Joback Method
tb	513.49	K	Joback Method
tc	690.21	K	Joback Method
tf	207.50 ± 0.10	K	NIST Webbook
vc	0.620	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	438.16	J/mol×K	631.30	Joback Method
cpg	450.23	J/mol×K	660.76	Joback Method
cpg	383.87	J/mol×K	513.49	Joback Method
cpg	398.38	J/mol×K	542.94	Joback Method
cpg	412.25	J/mol×K	572.40	Joback Method
cpg	425.51	J/mol×K	601.85	Joback Method
cpg	461.73	J/mol×K	690.21	Joback Method
hfust	13.08	kJ/mol	207.50	NIST Webbook
sfust	63.04	J/mol×K	207.50	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C105168&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C105168&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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