

# «gamma»-(4-Fluorophenyl)-«gamma»-butyrolactone

<b>Other names:</b>	«gamma»-(4-Fluorophenyl)-butyrolactone 2(3H)-Furanone, 5-(4-fluorophenyl)dihydro-
<b>Inchi:</b>	InChI=1S/C10H9FO2/c11-8-3-1-7(2-4-8)9-5-6-10(12)13-9/h1-4,9H,5-6H2
<b>InchiKey:</b>	RMFNZGXVLAUJHF-UHFFFAOYSA-N
<b>Formula:</b>	C10H9FO2
<b>SMILES:</b>	O=C1CCC(c2ccc(F)cc2)O1
<b>Mol. weight [g/mol]:</b>	180.18
<b>CAS:</b>	51787-96-3

## Physical Properties

Property code	Value	Unit	Source
gf	-230.87	kJ/mol	Joback Method
hf	-430.00	kJ/mol	Joback Method
hfus	19.81	kJ/mol	Joback Method
hvap	48.99	kJ/mol	Joback Method
log10ws	-2.66		Crippen Method
logp	2.204		Crippen Method
mcvol	126.350	ml/mol	McGowan Method
pc	3456.14	kPa	Joback Method
tb	569.18	K	Joback Method
tc	811.40	K	Joback Method
tf	347.68	K	Joback Method
vc	0.474	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	307.75	J/molxK	569.18	Joback Method
cpg	323.26	J/molxK	609.55	Joback Method
cpg	337.76	J/molxK	649.92	Joback Method
cpg	351.25	J/molxK	690.29	Joback Method
cpg	363.75	J/molxK	730.66	Joback Method
cpg	375.27	J/molxK	771.03	Joback Method
cpg	385.83	J/molxK	811.40	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C51787963&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C51787963&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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