

# Benzenepropanenitrile, «alpha»,3-dichloro-2-methyl-

Other names:	CCTP FMC 28979 Hydrocinnamonitrile, «alpha»,3-Dichloro-2-methyl- Orthonil PRB 8 PRB 200 2-Chloro-3-(3-chloro-o-tolyl)-propionitrile Bi-Act Propionitrile, 2-chloro-3-(3-chloro-o-tolyl)- NSC 239071 2-(2-Chloro-2-cyanoethyl)-6-chlorotolyl 2-chloro-3-(m-chloro-o-tolyl)propionitrile
Inchi:	InChI=1S/C10H9Cl2N/c1-7-8(5-9(11)6-13)3-2-4-10(7)12/h2-4,9H,5H2,1H3
InchiKey:	YYJUXSGXHHPBTK-UHFFFAOYSA-N
Formula:	C10H9Cl2N
SMILES:	<chem>Cc1c(Cl)cccc1CC(Cl)C#N</chem>
Mol. weight [g/mol]:	214.09
CAS:	21342-85-8

## Physical Properties

Property code	Value	Unit	Source
gf	233.35	kJ/mol	Joback Method
hf	91.98	kJ/mol	Joback Method
hfus	21.30	kJ/mol	Joback Method
hvap	60.31	kJ/mol	Joback Method
log10ws	-3.99		Crippen Method
logp	3.322		Crippen Method
mcvol	153.860	ml/mol	McGowan Method
pc	2581.96	kPa	Joback Method
tb	641.34	K	Joback Method
tc	878.81	K	Joback Method
tf	363.75	K	Joback Method
vc	0.606	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	333.07	J/mol×K	641.34	Joback Method
cpg	343.57	J/mol×K	680.92	Joback Method
cpg	353.34	J/mol×K	720.50	Joback Method
cpg	362.39	J/mol×K	760.08	Joback Method
cpg	370.78	J/mol×K	799.65	Joback Method
cpg	378.53	J/mol×K	839.23	Joback Method
cpg	385.67	J/mol×K	878.81	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	384.20	K	0.07	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C21342858&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C21342858&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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