

# o-Anisic acid, 3-chloroprop-2-enyl ester

<b>Inchi:</b>	InChI=1S/C11H11ClO3/c1-14-10-6-3-2-5-9(10)11(13)15-8-4-7-12/h2-7H,8H2,1H3/b7-4+
<b>InchiKey:</b>	WBWAKPXHPTXRTE-QPJJXVBHSA-N
<b>Formula:</b>	C11H11ClO3
<b>SMILES:</b>	COc1ccccc1C(=O)OCC=CCl
<b>Mol. weight [g/mol]:</b>	226.66

## Physical Properties

Property code	Value	Unit	Source
gf	-126.11	kJ/mol	Joback Method
hf	-320.85	kJ/mol	Joback Method
hfus	26.27	kJ/mol	Joback Method
hvap	58.93	kJ/mol	Joback Method
log10ws	-3.17		Crippen Method
logp	2.604		Crippen Method
mcvol	163.340	ml/mol	McGowan Method
pc	2729.71	kPa	Joback Method
rinpol	1730.00		NIST Webbook
rinpol	1730.00		NIST Webbook
tb	623.04	K	Joback Method
tc	843.89	K	Joback Method
tf	371.90	K	Joback Method
vc	0.615	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	375.97	J/molxK	623.04	Joback Method
cpg	430.65	J/molxK	807.08	Joback Method
cpg	421.20	J/molxK	770.27	Joback Method
cpg	411.03	J/molxK	733.47	Joback Method
cpg	400.11	J/molxK	696.66	Joback Method
cpg	388.43	J/molxK	659.85	Joback Method
cpg	439.39	J/molxK	843.89	Joback Method
dvisc	0.0001304	Paxs	623.04	Joback Method

dvisc	0.0001637	Paxs	581.18	Joback Method
dvisc	0.0002129	Paxs	539.33	Joback Method
dvisc	0.0002893	Paxs	497.47	Joback Method
dvisc	0.0004160	Paxs	455.61	Joback Method
dvisc	0.0006437	Paxs	413.76	Joback Method
dvisc	0.0010989	Paxs	371.90	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U292580&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U292580&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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