

Acetic acid, trifluoro-, 2-phenylethyl ester

Other names:	1-Trifluoroacetyloxy-2-phenylethane 2-Phenylethanol, trifluoroacetate
Inchi:	InChI=1S/C10H9F3O2/c11-10(12,13)9(14)15-7-6-8-4-2-1-3-5-8/h1-5H,6-7H2
InchiKey:	JQKSSRRXELCGQS-UHFFFAOYSA-N
Formula:	C10H9F3O2
SMILES:	O=C(OCCc1ccccc1)C(F)(F)F
Mol. weight [g/mol]:	218.17
CAS:	55419-66-4

Physical Properties

Property code	Value	Unit	Source
gf	-669.78	kJ/mol	Joback Method
hf	-855.08	kJ/mol	Joback Method
hfus	20.31	kJ/mol	Joback Method
hvap	45.54	kJ/mol	Joback Method
log10ws	-2.64		Crippen Method
logp	2.335		Crippen Method
mvol	140.750	ml/mol	McGowan Method
pc	2721.17	kPa	Joback Method
rinpol	1086.00		NIST Webbook
tb	525.75	K	Joback Method
tc	720.19	K	Joback Method
tf	305.23	K	Joback Method
vc	0.554	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	328.67	J/molxK	525.75	Joback Method
cpg	341.23	J/molxK	558.16	Joback Method
cpg	353.00	J/molxK	590.56	Joback Method
cpg	364.01	J/molxK	622.97	Joback Method
cpg	374.29	J/molxK	655.38	Joback Method
cpg	383.87	J/molxK	687.78	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C55419664&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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