

Sodium sulfite

Other names:	disodium sulfite sulfurous acid, sodium salt
Inchi:	InChI=1S/2Na.H2O3S/c;;1-4(2)3/h;;(H2,1,2,3)/q2*+1;/p-2
InchiKey:	GEHJYWRUCIMESM-UHFFFAOYSA-L
Formula:	Na2O3S
SMILES:	O=S(O[Na])O[Na]
Mol. weight [g/mol]:	126.04

Sources

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=B6000566&Units=SI>

Liquid liquid equilibrium of 1-propanol, 2-propanol, 2-methyl-2-propanol or <https://www.doi.org/10.1016/j.fluid.2012.05.012>

Salting-out effect of sodium, water <https://www.doi.org/10.1016/j.fluid.2013.09.028>

potassium carbonate, sulfite, tartrate <https://www.doi.org/10.1016/j.tca.2010.01.020>

and sodium sulfite on successive phase <https://www.doi.org/10.1021/acs.jced.7b01089>

systems containing trimethylborate <https://www.doi.org/10.1021/acs.jced.7b01089>

and methylamine in the presence of <https://www.doi.org/10.1021/je200379r>

solubility of sodium: Experimental <https://www.doi.org/10.1021/je200379r>

and sodium carbonate, sodium sulfite, <https://www.doi.org/10.1021/je200379r>

and sodium phosphate in the <https://www.doi.org/10.1021/je200379r>

presence of sodium chloride in <https://www.doi.org/10.1021/je200379r>

aqueous solutions of sodium <https://www.doi.org/10.1021/je200379r>

chloride and sodium <https://www.doi.org/10.1021/je200379r>

phosphate/sodium chloride/sodium <https://www.doi.org/10.1021/je200379r>

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