

# 2,4,6-Trichlorobenzyl alcohol, 2-methylpropyl ether

Inchi:	InChI=1S/C11H13Cl3O/c1-7(2)5-15-6-9-10(13)3-8(12)4-11(9)14/h3-4,7H,5-6H2,1-2H3
InchiKey:	DDQQMOKLWNGIOC-UHFFFAOYSA-N
Formula:	C11H13Cl3O
SMILES:	CC(C)COCc1c(Cl)cc(Cl)cc1Cl
Mol. weight [g/mol]:	267.58

## Physical Properties

Property code	Value	Unit	Source
gf	-17.97	kJ/mol	Joback Method
hf	-252.97	kJ/mol	Joback Method
hfus	27.38	kJ/mol	Joback Method
hvap	59.52	kJ/mol	Joback Method
log10ws	-4.93		Crippen Method
logp	4.819		Crippen Method
mcvol	184.680	ml/mol	McGowan Method
pc	2250.40	kPa	Joback Method
rinpol	1658.00		NIST Webbook
rinpol	1658.00		NIST Webbook
tb	626.97	K	Joback Method
tc	847.93	K	Joback Method
tf	374.70	K	Joback Method
vc	0.703	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	407.09	J/molxK	626.97	Joback Method
cpg	419.80	J/molxK	663.80	Joback Method
cpg	431.76	J/molxK	700.62	Joback Method
cpg	443.00	J/molxK	737.45	Joback Method
cpg	453.51	J/molxK	774.28	Joback Method
cpg	463.32	J/molxK	811.10	Joback Method
cpg	472.44	J/molxK	847.93	Joback Method
dvisc	0.0011413	Paxs	374.70	Joback Method

dvisc	0.0006843	Paxs	416.75	Joback Method
dvisc	0.0004506	Paxs	458.79	Joback Method
dvisc	0.0003183	Paxs	500.84	Joback Method
dvisc	0.0002373	Paxs	542.88	Joback Method
dvisc	0.0001845	Paxs	584.92	Joback Method
dvisc	0.0001484	Paxs	626.97	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375280&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375280&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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