

# Butanoic acid, 4-(2,4,5-trichlorophenoxy)-

**Other names:**

Butyric acid, 4-(2,4,5-trichlorophenoxy)-  
«gamma»-(2,4,5-Trichlorophenoxy)butyric acid  
2,4,5-TB  
4-(2,4,5-Trichlorophenoxy)Butyric acid  
2,4,5-T Butyric acid  
4-(2,4,5-Trichlorophenoxy)butanoic acid

**Inchi:**

InChI=1S/C10H9Cl3O3/c11-6-4-8(13)9(5-7(6)12)16-3-1-2-10(14)15/h4-5H,1-3H2,(H,14,15)

**InchiKey:**

RTWCZQFXFMXXKP-UHFFFAOYSA-N

**Formula:**

C10H9Cl3O3

**SMILES:**

O=C(O)CCCOc1cc(Cl)c(Cl)cc1Cl

**Mol. weight [g/mol]:**

283.54

**CAS:**

93-80-1

## Physical Properties

Property code	Value	Unit	Source
gf	-289.69	kJ/mol	Joback Method
hf	-491.86	kJ/mol	Joback Method
hfus	34.00	kJ/mol	Joback Method
hvap	81.11	kJ/mol	Joback Method
log10ws	-4.00		Crippen Method
logp	3.890		Crippen Method
mcvol	178.030	ml/mol	McGowan Method
pc	2875.03	kPa	Joback Method
tb	750.58	K	Joback Method
tc	962.26	K	Joback Method
tf	386.80 ± 0.20	K	NIST Webbook
tf	388.10 ± 0.20	K	NIST Webbook
vc	0.677	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	453.38	J/mol×K	962.26	Joback Method
cpg	412.17	J/mol×K	750.58	Joback Method

cpg	420.49	J/mol×K	785.86	Joback Method
cpg	428.22	J/mol×K	821.14	Joback Method
cpg	435.36	J/mol×K	856.42	Joback Method
cpg	441.93	J/mol×K	891.70	Joback Method
cpg	447.93	J/mol×K	926.98	Joback Method
dvisc	0.0000384	Paxs	750.58	Joback Method
dvisc	0.0005473	Paxs	489.18	Joback Method
dvisc	0.0002932	Paxs	532.75	Joback Method
dvisc	0.0001726	Paxs	576.31	Joback Method
dvisc	0.0001095	Paxs	619.88	Joback Method
dvisc	0.0000737	Paxs	663.45	Joback Method
dvisc	0.0000521	Paxs	707.01	Joback Method
hfust	30.28	kJ/mol	386.70	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C93801&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C93801&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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