

# 4-Nitrobenzoic acid, 2-pentyl ester

<b>Other names:</b>	Benzoic acid, 4-nitro, 1-methylbutyl ester
<b>Inchi:</b>	InChI=1S/C12H15NO4/c1-3-4-9(2)17-12(14)10-5-7-11(8-6-10)13(15)16/h5-9H,3-4H2,1-2
<b>InchiKey:</b>	BAJPTXADVLNCKW-UHFFFAOYSA-N
<b>Formula:</b>	C12H15NO4
<b>SMILES:</b>	CCCC(C)OC(=O)c1ccc([N+](=O)[O-])cc1
<b>Mol. weight [g/mol]:</b>	237.25
<b>CAS:</b>	100391-69-3

## Physical Properties

Property code	Value	Unit	Source
gf	-47.87	kJ/mol	Joback Method
hf	-326.79	kJ/mol	Joback Method
hfus	31.11	kJ/mol	Joback Method
hvap	70.60	kJ/mol	Joback Method
log10ws	-4.15		Crippen Method
logp	2.940		Crippen Method
mcvol	181.040	ml/mol	McGowan Method
pc	2530.27	kPa	Joback Method
rinpol	1721.00		NIST Webbook
rinpol	1721.00		NIST Webbook
rinpol	1724.00		NIST Webbook
rinpol	1740.00		NIST Webbook
rinpol	1724.00		NIST Webbook
rinpol	1724.00		NIST Webbook
ripol	2448.00		NIST Webbook
ripol	2438.00		NIST Webbook
ripol	2438.00		NIST Webbook
ripol	2490.00		NIST Webbook
ripol	2463.00		NIST Webbook
ripol	2448.00		NIST Webbook
tb	733.31	K	Joback Method
tc	966.01	K	Joback Method
tf	464.71	K	Joback Method
vc	0.700	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	498.12	J/molxK	733.31	Joback Method
cpg	511.61	J/molxK	772.09	Joback Method
cpg	524.08	J/molxK	810.88	Joback Method
cpg	535.55	J/molxK	849.66	Joback Method
cpg	546.05	J/molxK	888.45	Joback Method
cpg	555.62	J/molxK	927.23	Joback Method
cpg	564.27	J/molxK	966.01	Joback Method

## Sources

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C100391693&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/91-263-9/4-Nitrobenzoic-acid-2-pentyl-ester.pdf>

Generated by Cheméo on 2024-04-19 14:44:25.514803304 +0000 UTC m=+15827114.435380614.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.