

# Glycine, n-(phenoxyacetyl)-, ethyl ester

Inchi:	InChI=1S/C12H15NO4/c1-2-16-12(15)8-13-11(14)9-17-10-6-4-3-5-7-10/h3-7H,2,8-9H2,1
InchiKey:	RUEKMCCFOHLOMQ-UHFFFAOYSA-N
Formula:	C12H15NO4
SMILES:	CCOC(=O)CN=C(O)COc1ccccc1
Mol. weight [g/mol]:	237.25
CAS:	27127-58-8

## Physical Properties

Property code	Value	Unit	Source
hf	-511.30	kJ/mol	Joback Method
hvap	76.22	kJ/mol	Joback Method
log10ws	-1.53		Crippen Method
logp	1.585		Crippen Method
mcvol	181.040	ml/mol	McGowan Method
pc	2455.60	kPa	Joback Method
tb	768.09	K	Joback Method
tc	976.34	K	Joback Method

## Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C27127588&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C27127588&amp;Units=SI</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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