

# Isopropyl 3-chloro-6-hydroxy-4-methoxy-2-methylbenzoate

<b>Inchi:</b>	InChI=1S/C12H15ClO4/c1-6(2)17-12(15)10-7(3)11(13)9(16-4)5-8(10)14/h5-6,14H,1-4H3
<b>InchiKey:</b>	KJHOFASQYUCZPN-UHFFFAOYSA-N
<b>Formula:</b>	C12H15ClO4
<b>SMILES:</b>	COc1cc(O)c(C(=O)OC(C)C)c(C)c1Cl
<b>Mol. weight [g/mol]:</b>	258.70

## Physical Properties

Property code	Value	Unit	Source
gf	-374.23	kJ/mol	Joback Method
hf	-664.24	kJ/mol	Joback Method
hfus	30.14	kJ/mol	Joback Method
hvap	75.14	kJ/mol	Joback Method
log10ws	-3.49		Crippen Method
logp	2.928		Crippen Method
mcvol	187.600	ml/mol	McGowan Method
pc	2670.78	kPa	Joback Method
rinpol	1904.80		NIST Webbook
rinpol	1904.80		NIST Webbook
tb	731.90	K	Joback Method
tc	954.64	K	Joback Method
tf	510.01	K	Joback Method
vc	0.650	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	494.23	J/molxK	731.90	Joback Method
cpg	506.47	J/molxK	769.02	Joback Method
cpg	517.99	J/molxK	806.15	Joback Method
cpg	528.85	J/molxK	843.27	Joback Method
cpg	539.07	J/molxK	880.39	Joback Method
cpg	548.68	J/molxK	917.52	Joback Method
cpg	557.71	J/molxK	954.64	Joback Method
dvisc	0.0001447	Paxs	510.01	Joback Method

dvisc	0.0000794	Paxs	546.99	Joback Method
dvisc	0.0000470	Paxs	583.97	Joback Method
dvisc	0.0000296	Paxs	620.96	Joback Method
dvisc	0.0000196	Paxs	657.94	Joback Method
dvisc	0.0000136	Paxs	694.92	Joback Method
dvisc	0.0000098	Paxs	731.90	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U413423&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U413423&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/91-270-1/Isopropyl-3-chloro-6-hydroxy-4-methoxy-2-methylbenzoate.pdf>

Generated by Cheméo on 2024-05-22 02:54:20.303878714 +0000 UTC m=+18635709.224456038.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.