

3-Methylbutan-2-yl 3-chlorobenzoate

Other names:	Benzoic acid, 3-chloro, 1,2-dimethylpropyl ester
Inchi:	InChI=1S/C12H15ClO2/c1-8(2)9(3)15-12(14)10-5-4-6-11(13)7-10/h4-9H,1-3H3
InchiKey:	UJFYEUKAHUQXLE-UHFFFAOYSA-N
Formula:	C12H15ClO2
SMILES:	CC(C)C(C)OC(=O)c1cccc(Cl)c1
Mol. weight [g/mol]:	226.70

Physical Properties

Property code	Value	Unit	Source
gf	-97.79	kJ/mol	Joback Method
hf	-337.05	kJ/mol	Joback Method
hfus	20.43	kJ/mol	Joback Method
hvap	58.01	kJ/mol	Joback Method
log10ws	-3.94		Crippen Method
logp	3.541		Crippen Method
mcvol	175.860	ml/mol	McGowan Method
pc	2429.05	kPa	Joback Method
rinpol	1531.00		NIST Webbook
rinpol	1551.00		NIST Webbook
rinpol	1527.00		NIST Webbook
rinpol	1512.00		NIST Webbook
rinpol	1521.00		NIST Webbook
rinpol	1551.00		NIST Webbook
rinpol	1514.00		NIST Webbook
rinpol	1523.00		NIST Webbook
rinpol	1514.00		NIST Webbook
ripol	1988.00		NIST Webbook
ripol	2010.00		NIST Webbook
ripol	1989.00		NIST Webbook
ripol	2023.00		NIST Webbook
ripol	2023.00		NIST Webbook
ripol	2007.00		NIST Webbook
ripol	2023.00		NIST Webbook
ripol	1988.00		NIST Webbook
tb	618.46	K	Joback Method
tc	838.01	K	Joback Method
tf	336.02	K	Joback Method

vc	0.660	m3/kmol	Joback Method
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Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	422.24	J/mol×K	618.46	Joback Method
cpg	436.93	J/mol×K	655.05	Joback Method
cpg	450.70	J/mol×K	691.64	Joback Method
cpg	463.59	J/mol×K	728.23	Joback Method
cpg	475.62	J/mol×K	764.82	Joback Method
cpg	486.79	J/mol×K	801.42	Joback Method
cpg	497.15	J/mol×K	838.01	Joback Method
dvisc	0.0024032	Pa×s	336.02	Joback Method
dvisc	0.0011407	Pa×s	383.09	Joback Method
dvisc	0.0006373	Pa×s	430.17	Joback Method
dvisc	0.0003994	Pa×s	477.24	Joback Method
dvisc	0.0002722	Pa×s	524.31	Joback Method
dvisc	0.0001976	Pa×s	571.39	Joback Method
dvisc	0.0001507	Pa×s	618.46	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U373557&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀ws:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
ri_{pol}:	Non-polar retention indices
ri_{pol}:	Polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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