

# Benzene, 1,3,5-tris(chloromethyl)-2,4,6-trimethyl-

Other names:	1,3,5-Trimethyl-2,4,6-tris(chloromethyl)benzene
Inchi:	InChI=1S/C12H15Cl3/c1-7-10(4-13)8(2)12(6-15)9(3)11(7)5-14/h4-6H2,1-3H3
InchiKey:	PHQFMPNZCIHSPC-UHFFFAOYSA-N
Formula:	C12H15Cl3
SMILES:	Cc1c(CCl)c(C)c(CCl)c(C)c1CCl
Mol. weight [g/mol]:	265.61
CAS:	3849-01-2

## Physical Properties

Property code	Value	Unit	Source
gf	78.63	kJ/mol	Joback Method
hf	-159.05	kJ/mol	Joback Method
hfus	31.52	kJ/mol	Joback Method
hvap	61.05	kJ/mol	Joback Method
log10ws	-6.00		Crippen Method
logp	4.828		Crippen Method
mcvol	192.900	ml/mol	McGowan Method
pc	2019.95	kPa	Joback Method
rinpol	2073.00		NIST Webbook
rinpol	2073.00		NIST Webbook
tb	637.83	K	Joback Method
tc	855.06	K	Joback Method
tf	403.78	K	Joback Method
vc	0.747	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	429.11	J/molxK	637.83	Joback Method
cpg	442.08	J/molxK	674.04	Joback Method
cpg	454.35	J/molxK	710.24	Joback Method
cpg	465.94	J/molxK	746.45	Joback Method
cpg	476.87	J/molxK	782.65	Joback Method
cpg	487.15	J/molxK	818.86	Joback Method

cpg	496.81	J/molxK	855.06	Joback Method
dvisc	0.0008623	Paxs	403.78	Joback Method
dvisc	0.0005910	Paxs	442.79	Joback Method
dvisc	0.0004306	Paxs	481.80	Joback Method
dvisc	0.0003290	Paxs	520.81	Joback Method
dvisc	0.0002610	Paxs	559.81	Joback Method
dvisc	0.0002134	Paxs	598.82	Joback Method
dvisc	0.0001788	Paxs	637.83	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3849012&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3849012&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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