

# Glutaric acid, but-3-en-2-yl 3-chlorophenyl ester

Inchi:	InChI=1S/C15H17ClO4/c1-3-11(2)19-14(17)8-5-9-15(18)20-13-7-4-6-12(16)10-13/h3-4,6
InchiKey:	AKESMPCJYGAJAN-UHFFFAOYSA-N
Formula:	C15H17ClO4
SMILES:	<chem>C=CC(C)OC(=O)CCCC(=O)Oc1cccc(Cl)c1</chem>
Mol. weight [g/mol]:	296.75

## Physical Properties

Property code	Value	Unit	Source
gf	-216.17	kJ/mol	Joback Method
hf	-513.06	kJ/mol	Joback Method
hfus	33.23	kJ/mol	Joback Method
hvap	73.56	kJ/mol	Joback Method
log10ws	-4.23		Crippen Method
logp	3.533		Crippen Method
mvol	221.270	ml/mol	McGowan Method
pc	1998.33	kPa	Joback Method
rinpol	2025.00		NIST Webbook
rinpol	2025.00		NIST Webbook
tb	760.51	K	Joback Method
tc	973.94	K	Joback Method
tf	455.23	K	Joback Method
vc	0.840	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	595.45	J/molxK	760.51	Joback Method
cpg	608.79	J/molxK	796.08	Joback Method
cpg	621.15	J/molxK	831.65	Joback Method
cpg	632.56	J/molxK	867.23	Joback Method
cpg	643.02	J/molxK	902.80	Joback Method
cpg	652.56	J/molxK	938.37	Joback Method
cpg	661.20	J/molxK	973.94	Joback Method
dvisc	0.0009033	Paxs	455.23	Joback Method

dvisc	0.0005103	Paxs	506.11	Joback Method
dvisc	0.0003200	Paxs	556.99	Joback Method
dvisc	0.0002170	Paxs	607.87	Joback Method
dvisc	0.0001562	Paxs	658.75	Joback Method
dvisc	0.0001179	Paxs	709.63	Joback Method
dvisc	0.0000924	Paxs	760.51	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405242&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405242&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/91-316-0/Glutaric-acid-but-3-en-2-yl-3-chlorophenyl-ester.pdf>

Generated by Cheméo on 2024-05-06 17:07:37.118896931 +0000 UTC m=+17304506.039474242.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.