

# pentyl pentafluorobenzoate

<b>Inchi:</b>	InChI=1S/C12H11F5O2/c1-2-3-4-5-19-12(18)6-7(13)9(15)11(17)10(16)8(6)14/h2-5H2,1H
<b>InchiKey:</b>	OWMXFTXURJTYJD-UHFFFAOYSA-N
<b>Formula:</b>	C12H11F5O2
<b>SMILES:</b>	CCCCCOC(=O)c1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	282.21

## Physical Properties

Property code	Value	Unit	Source
gf	-1093.55	kJ/mol	Joback Method
hf	-1337.18	kJ/mol	Joback Method
hfus	37.12	kJ/mol	Joback Method
hvap	52.96	kJ/mol	Joback Method
log10ws	-5.05		Crippen Method
logp	3.729		Crippen Method
mcvol	172.470	ml/mol	McGowan Method
pc	1901.92	kPa	Joback Method
ripol	1340.00		NIST Webbook
ripol	1328.00		NIST Webbook
ripol	1322.00		NIST Webbook
ripol	1310.00		NIST Webbook
ripol	1322.00		NIST Webbook
ripol	1322.00		NIST Webbook
ripol	1319.00		NIST Webbook
ripol	1321.00		NIST Webbook
ripol	1319.00		NIST Webbook
ripol	1608.00		NIST Webbook
ripol	1632.00		NIST Webbook
ripol	1635.00		NIST Webbook
ripol	1607.00		NIST Webbook
ripol	1632.00		NIST Webbook
ripol	1564.00		NIST Webbook
ripol	1595.00		NIST Webbook
ripol	1611.00		NIST Webbook
ripol	1608.00		NIST Webbook
tb	598.18	K	Joback Method
tc	768.75	K	Joback Method
tf	389.13	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	432.76	J/mol×K	598.18	Joback Method
cpg	443.90	J/mol×K	626.61	Joback Method
cpg	454.59	J/mol×K	655.04	Joback Method
cpg	464.81	J/mol×K	683.47	Joback Method
cpg	474.58	J/mol×K	711.90	Joback Method
cpg	483.89	J/mol×K	740.33	Joback Method
cpg	492.74	J/mol×K	768.75	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R311965&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R311965&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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