

p-mentha-2,8-dien-1-yl acetate

Inchi:	InChI=1S/C12H18O2/c1-9(2)11-5-7-12(4,8-6-11)14-10(3)13/h5,7,11H,1,6,8H2,2-4H3
InchiKey:	UYKRLPCOIHAEFQ-UHFFFAOYSA-N
Formula:	C12H18O2
SMILES:	<chem>C=C(C)C1C=CC(C)(OC(C)=O)CC1</chem>
Mol. weight [g/mol]:	194.27

Physical Properties

Property code	Value	Unit	Source
gf	-63.26	kJ/mol	Joback Method
hf	-313.17	kJ/mol	Joback Method
hfus	14.86	kJ/mol	Joback Method
hvap	50.13	kJ/mol	Joback Method
log10ws	-3.18		Crippen Method
logp	2.850		Crippen Method
mcvol	167.920	ml/mol	McGowan Method
pc	2441.06	kPa	Joback Method
ripol	1731.00		NIST Webbook
tb	561.09	K	Joback Method
tc	778.26	K	Joback Method
tf	309.24	K	Joback Method
vc	0.629	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	414.24	J/molxK	561.09	Joback Method
cpg	432.26	J/molxK	597.29	Joback Method
cpg	449.21	J/molxK	633.48	Joback Method
cpg	465.20	J/molxK	669.68	Joback Method
cpg	480.34	J/molxK	705.87	Joback Method
cpg	494.73	J/molxK	742.07	Joback Method
cpg	508.47	J/molxK	778.26	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R320533&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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