

# Succinic acid, monochloride 2-methylpent-3-yl ester

<b>Inchi:</b>	InChI=1S/C10H17ClO3/c1-4-8(7(2)3)14-10(13)6-5-9(11)12/h7-8H,4-6H2,1-3H3
<b>InchiKey:</b>	LPYUZXWODZHGIK-UHFFFAOYSA-N
<b>Formula:</b>	C10H17ClO3
<b>SMILES:</b>	CCC(OC(=O)CCC(=O)Cl)C(C)C
<b>Mol. weight [g/mol]:</b>	220.69

## Physical Properties

Property code	Value	Unit	Source
gf	-346.33	kJ/mol	Joback Method
hf	-633.41	kJ/mol	Joback Method
hfus	23.19	kJ/mol	Joback Method
hvap	57.36	kJ/mol	Joback Method
log10ws	-2.67		Crippen Method
logp	2.510		Crippen Method
mcvol	173.010	ml/mol	McGowan Method
pc	2289.32	kPa	Joback Method
rinpol	1367.00		NIST Webbook
rinpol	1367.00		NIST Webbook
tb	594.91	K	Joback Method
tc	787.34	K	Joback Method
tf	324.47	K	Joback Method
vc	0.662	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	419.12	J/mol×K	594.91	Joback Method
cpg	479.10	J/mol×K	755.26	Joback Method
cpg	468.38	J/mol×K	723.19	Joback Method
cpg	457.03	J/mol×K	691.12	Joback Method
cpg	445.04	J/mol×K	659.05	Joback Method
cpg	432.41	J/mol×K	626.98	Joback Method
cpg	489.20	J/mol×K	787.34	Joback Method
dvisc	0.0001969	Paxs	594.91	Joback Method

dvisc	0.0002635	Paxs	549.84	Joback Method
dvisc	0.0003714	Paxs	504.76	Joback Method
dvisc	0.0005600	Paxs	459.69	Joback Method
dvisc	0.0009231	Paxs	414.62	Joback Method
dvisc	0.0017189	Paxs	369.54	Joback Method
dvisc	0.0038044	Paxs	324.47	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U349401&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U349401&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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