

1,2-Diaminopropane

Inchi:	InChI=1S/C3H10N2/c1-3(5)2-4/h3H,2,4-5H2,1H3
InchiKey:	AOHJOMMDDJHIJH-UHFFFAOYSA-N
Formula:	C3H10N2
SMILES:	CC(N)CN
Mol. weight [g/mol]:	74.12
CAS:	10424-38-1

Physical Properties

Property code	Value	Unit	Source
gf	104.84	kJ/mol	Joback Method
hf	-53.68 ± 0.46	kJ/mol	NIST Webbook
hfus	10.40	kJ/mol	Joback Method
hvap	44.12 ± 0.21	kJ/mol	NIST Webbook
log10ws	-0.06		Crippen Method
logp	-0.708		Crippen Method
mcvol	73.090	ml/mol	McGowan Method
pc	5220.69	kPa	Joback Method
tb	392.70	K	NIST Webbook
tc	616.86	K	Joback Method
tf	275.09	K	Joback Method
vc	0.256	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	148.34	J/mol×K	412.66	Joback Method
cpg	156.74	J/mol×K	446.69	Joback Method
cpg	164.75	J/mol×K	480.73	Joback Method
cpg	172.39	J/mol×K	514.76	Joback Method
cpg	179.66	J/mol×K	548.79	Joback Method
cpg	186.57	J/mol×K	582.83	Joback Method
cpg	193.14	J/mol×K	616.86	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C10424381&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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