

# Pimelic acid, 3,7-dimethyloctyl 4-methyl-2-pentyl ester

<b>Inchi:</b>	InChI=1S/C23H44O4/c1-18(2)11-10-12-20(5)15-16-26-22(24)13-8-7-9-14-23(25)27-21(6)
<b>InchiKey:</b>	TTZQTCFUANRIIV-UHFFFAOYSA-N
<b>Formula:</b>	C23H44O4
<b>SMILES:</b>	CC(C)CCCC(C)CCOC(=O)CCCCC(=O)OC(C)CC(C)C
<b>Mol. weight [g/mol]:</b>	384.59

## Physical Properties

Property code	Value	Unit	Source
gf	-334.82	kJ/mol	Joback Method
hf	-1028.77	kJ/mol	Joback Method
hfus	46.81	kJ/mol	Joback Method
hvap	83.55	kJ/mol	Joback Method
log10ws	-6.56		Crippen Method
logp	6.310		Crippen Method
mvol	349.810	ml/mol	McGowan Method
pc	917.16	kPa	Joback Method
rinpol	2424.00		NIST Webbook
rinpol	2424.00		NIST Webbook
tb	876.46	K	Joback Method
tc	1073.32	K	Joback Method
tf	433.29	K	Joback Method
vc	1.347	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1156.28	J/molxK	876.46	Joback Method
cpg	1175.85	J/molxK	909.27	Joback Method
cpg	1194.08	J/molxK	942.08	Joback Method
cpg	1211.02	J/molxK	974.89	Joback Method
cpg	1226.67	J/molxK	1007.70	Joback Method
cpg	1241.08	J/molxK	1040.51	Joback Method
cpg	1254.27	J/molxK	1073.32	Joback Method
dvisc	0.0012150	Paxs	433.29	Joback Method

dvisc	0.0003922	Paxs	507.15	Joback Method
dvisc	0.0001688	Paxs	581.01	Joback Method
dvisc	0.0000878	Paxs	654.88	Joback Method
dvisc	0.0000522	Paxs	728.74	Joback Method
dvisc	0.0000341	Paxs	802.60	Joback Method
dvisc	0.0000240	Paxs	876.46	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393738&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393738&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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