

4-Methyl-1,2,3-trithiolane

Inchi:	InChI=1S/C3H6S3/c1-3-2-4-6-5-3/h3H,2H2,1H3
InchiKey:	MBYUXUPOTRBMDQ-UHFFFAOYSA-N
Formula:	C3H6S3
SMILES:	CC1CSSS1
Mol. weight [g/mol]:	138.28

Physical Properties

Property code	Value	Unit	Source
hf	52.82	kJ/mol	Cheméo Relay (1.0)
hfus	8.43	kJ/mol	Joback Method
ie	8.42	eV	Cheméo Relay (1.0)
logp	2.418		Crippen Method
mcvol	91.320	ml/mol	McGowan Method
rinpol	1157.30		NIST Webbook
tb	494.61	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	151.28	J/mol×K	426.81	Joback Method
cpg	161.14	J/mol×K	469.90	Joback Method
cpg	170.27	J/mol×K	512.99	Joback Method
cpg	178.72	J/mol×K	556.08	Joback Method
cpg	186.53	J/mol×K	599.17	Joback Method
cpg	193.74	J/mol×K	642.26	Joback Method
cpg	200.40	J/mol×K	685.35	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C116664290&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>
Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>
Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):
Cheméo Relay (1.0, beta):

Legend

cpg:	Ideal gas heat capacity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
ie:	Ionization energy
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature

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