

# 5-(((1R,3aR,4S,6aR)-4-(Benzo[d][1,3]dioxol-5-yl)he

<b>Inchi:</b>	InChI=1S/C20H18O7/c1-3-15-17(25-9-23-15)5-11(1)19-13-7-22-20(14(13)8-21-19)27-12
<b>InchiKey:</b>	ZZMNWJVJUKMZJY-UHFFFAOYSA-N
<b>Formula:</b>	C20H18O7
<b>SMILES:</b>	c1cc2c(cc1OC1OCC3C(c4ccc5c(c4)OCO5)OCC13)OCO2
<b>Mol. weight [g/mol]:</b>	370.35
<b>CAS:</b>	526-07-8

## Physical Properties

Property code	Value	Unit	Source
gf	-99.10	kJ/mol	Joback Method
hf	-674.29	kJ/mol	Joback Method
hfus	71.48	kJ/mol	Joback Method
hvap	96.78	kJ/mol	Joback Method
log10ws	-4.27		Crippen Method
logp	2.883		Crippen Method
mvol	242.790	ml/mol	McGowan Method
pc	2410.00	kPa	Joback Method
rinpol	3208.40		NIST Webbook
rinpol	3208.40		NIST Webbook
tb	949.90	K	Joback Method
tc	1212.72	K	Joback Method
tf	664.45	K	Joback Method
vc	0.895	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	842.77	J/molxK	949.90	Joback Method
cpg	911.82	J/molxK	1168.92	Joback Method
cpg	898.98	J/molxK	1125.11	Joback Method
cpg	885.86	J/molxK	1081.31	Joback Method
cpg	872.26	J/molxK	1037.51	Joback Method
cpg	857.96	J/molxK	993.70	Joback Method
cpg	924.58	J/molxK	1212.72	Joback Method

dvisc	0.0018052	Paxs	949.90	Joback Method
dvisc	0.0020062	Paxs	902.32	Joback Method
dvisc	0.0022560	Paxs	854.75	Joback Method
dvisc	0.0025722	Paxs	807.17	Joback Method
dvisc	0.0029813	Paxs	759.60	Joback Method
dvisc	0.0035242	Paxs	712.02	Joback Method
dvisc	0.0042671	Paxs	664.45	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C526078&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C526078&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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