

# Dimethyl 3-hydroxy-3-methylpentane-1,5-dioate

<b>Other names:</b>	Pentanedioic acid, 3-hydroxy-3-methyl-, dimethyl ester dimethyl 3-hydroxy-3-methylglutarate
<b>Inchi:</b>	InChI=1S/C8H14O5/c1-8(11,4-6(9)12-2)5-7(10)13-3/h11H,4-5H2,1-3H3
<b>InchiKey:</b>	ANEMYUYZEOZEJW-UHFFFAOYSA-N
<b>Formula:</b>	C8H14O5
<b>SMILES:</b>	COC(=O)CC(C)(O)CC(=O)OC
<b>Mol. weight [g/mol]:</b>	190.19
<b>CAS:</b>	56652-39-2

## Physical Properties

Property code	Value	Unit	Source
gf	-585.34	kJ/mol	Joback Method
hf	-859.03	kJ/mol	Joback Method
hfus	18.72	kJ/mol	Joback Method
hvap	67.10	kJ/mol	Joback Method
log10ws	-0.27		Crippen Method
logp	-0.136		Crippen Method
mcvol	144.330	ml/mol	McGowan Method
pc	3145.56	kPa	Joback Method
rinpol	1191.00		NIST Webbook
rinpol	1191.00		NIST Webbook
rinpol	1191.00		NIST Webbook
tb	623.97	K	Joback Method
tc	808.40	K	Joback Method
tf	387.48	K	Joback Method
vc	0.539	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	376.01	J/molxK	623.97	Joback Method
cpg	386.31	J/molxK	654.71	Joback Method
cpg	396.07	J/molxK	685.45	Joback Method
cpg	405.32	J/molxK	716.18	Joback Method

cpg	414.05	J/molxK	746.92	Joback Method
cpg	422.28	J/molxK	777.66	Joback Method
cpg	430.00	J/molxK	808.40	Joback Method
dvisc	0.0024147	Paxs	387.48	Joback Method
dvisc	0.0009857	Paxs	426.90	Joback Method
dvisc	0.0004681	Paxs	466.31	Joback Method
dvisc	0.0002497	Paxs	505.73	Joback Method
dvisc	0.0001459	Paxs	545.14	Joback Method
dvisc	0.0000916	Paxs	584.56	Joback Method
dvisc	0.0000610	Paxs	623.97	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C56652392&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C56652392&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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