

# 1-Hexanol, 6-amino-

<b>Other names:</b>	1-Amino-6-hexanol 6-Amino-1-hexanol 6-Aminohexanol 6-Aminohexyl alcohol 6-Hydroxyhexylamine 6-aminohexan-1-ol Amidohexylalkohol Aminohexyl alcohol H2NCH2(CH2)4CH2OH Hexanol, 6-amino- NH2(CH2)6OH NSC 91538
<b>Inchi:</b>	InChI=1S/C6H15NO/c7-5-3-1-2-4-6-8/h8H,1-7H2
<b>InchiKey:</b>	SUTWPJHCRAITLU-UHFFFAOYSA-N
<b>Formula:</b>	C6H15NO
<b>SMILES:</b>	NCCCCCO
<b>Mol. weight [g/mol]:</b>	117.19
<b>CAS:</b>	4048-33-3

## Physical Properties

Property code	Value	Unit	Source
affp	969.00	kJ/mol	NIST Webbook
basg	915.70	kJ/mol	NIST Webbook
gf	-70.73	kJ/mol	Joback Method
hf	-285.61	kJ/mol	Joback Method
hfus	20.58	kJ/mol	Joback Method
hvap	56.27	kJ/mol	Joback Method
log10ws	-1.03		Crippen Method
logp	0.498		Crippen Method
mcvol	111.250	ml/mol	McGowan Method
pc	3695.46	kPa	Joback Method
rinpol	1094.00		NIST Webbook
rinpol	1094.00		NIST Webbook
rinpol	1156.00		NIST Webbook
ripol	1902.00		NIST Webbook
ripol	1902.00		NIST Webbook
tb	501.39	K	Joback Method

tc	674.34	K	Joback Method
tf	330.00 ± 1.00	K	NIST Webbook
vc	0.419	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	260.58	J/mol×K	501.39	Joback Method
cpg	270.56	J/mol×K	530.21	Joback Method
cpg	280.12	J/mol×K	559.04	Joback Method
cpg	289.29	J/mol×K	587.86	Joback Method
cpg	298.06	J/mol×K	616.69	Joback Method
cpg	306.46	J/mol×K	645.51	Joback Method
cpg	314.50	J/mol×K	674.34	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	410.70	K	4.00	NIST Webbook
tbrp	410.50 ± 2.50	K	4.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.76652e+01
Coeff. B	-5.35334e+03
Coeff. C	-8.18600e+01
Temperature range (K), min.	389.92
Temperature range (K), max.	515.20

# Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4048333&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4048333&amp;Units=SI</a>

# Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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