(S)-Isopropyl lactate

Other names: isopropyl (S)-(-)-lactate

isopropyl (S)-2-hydroxypropionate

isopropyl (S)-lactate isopropyl S-(-)-lactate

propanoic acid, 2-hydroxy-, 1-methylethyl ester, (S)-

InChl=1S/C6H12O3/c1-4(2)9-6(8)5(3)7/h4-5,7H,1-3H3/t5-/m1/s1

InchiKey: KIWATKANDHUUOB-RXMQYKEDSA-N

Formula: C6H12O3

SMILES: CC(C)OC(=O)C(C)O

Mol. weight [g/mol]: 132.16 **CAS:** 63697-00-7

Physical Properties

Property code	Value	Unit	Source
gf	-375.98	kJ/mol	Joback Method
hf	-574.76	kJ/mol	Joback Method
hfus	11.12	kJ/mol	Joback Method
hvap	54.01	kJ/mol	Joback Method
log10ws	-0.68		Crippen Method
logp	0.319		Crippen Method
mcvol	108.710	ml/mol	McGowan Method
рс	3718.02	kPa	Joback Method
tb	504.27	K	Joback Method
tc	682.14	K	Joback Method
tf	260.36	K	Joback Method
VC	0.403	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	296.77	J/mol×K	682.14	Joback Method
cpg	246.04	J/mol×K	504.27	Joback Method
cpg	255.38	J/mol×K	533.92	Joback Method
cpg	264.36	J/mol×K	563.56	Joback Method

cpg	272.99	J/mol×K	593.21	Joback Method	
cpg	281.27	J/mol×K	622.85	Joback Method	
cpg	289.19	J/mol×K	652.50	Joback Method	
dvisc	0.0001409	Paxs	504.27	Joback Method	
dvisc	0.0398736	Paxs	260.36	Joback Method	
dvisc	0.0082434	Paxs	301.01	Joback Method	
dvisc	0.0024799	Paxs	341.66	Joback Method	
dvisc	0.0009632	Paxs	382.31	Joback Method	
dvisc	0.0004487	Paxs	422.97	Joback Method	
dvisc	0.0002390	Paxs	463.62	Joback Method	
pvap	0.41	kPa	306.80	Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods	
pvap	0.09	kPa	283.40	Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods	
pvap	0.12	kPa	288.30	Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods	
pvap	0.18	kPa	293.30	Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods	

pvap	0.24	kPa	298.30	Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods	
pvap	0.32	kPa	303.20	Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods	
rfi	1.41050		293.15	Vapor liquid equilibria and excess volumes of the binary systems ethanol + ethyl lactate, isopropanol + isopropyl lactate and n-butanol + n-butyl lactate at 101.325 kPa	

Sources

Joback Method: https://en.wikipedia.org/wiki/Joback_method

McGowan Method: http://link.springer.com/article/10.1007/BF02311772

NIST Webbook: http://webbook.nist.gov/cgi/cbook.cgi?ID=C63697007&Units=SI

Crippen Method: http://pubs.acs.org/doi/abs/10.1021/ci990307l

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Vapor liquid equilibria and excess volumes of the binary systems ethanol Remy abstaclasticoprobational assopropyl Example alternational experimental and computational methods:

https://www.doi.org/10.1016/j.fluid.2005.02.015 https://www.doi.org/10.1016/j.jct.2018.07.029

Legend

cpg: Ideal gas heat capacitydvisc: Dynamic viscosity

gf: Standard Gibbs free energy of formation

hf: Enthalpy of formation at standard conditionshfus: Enthalpy of fusion at standard conditions

hvap: Enthalpy of vaporization at standard conditions

log10ws: Log10 of Water solubility in mol/llogp: Octanol/Water partition coefficientmcvol: McGowan's characteristic volume

pc: Critical Pressurepvap: Vapor pressurerfi: Refractive Index

tb: Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) point

vc: Critical Volume

Latest version available from:

https://www.chemeo.com/cid/91-934-4/S-Isopropyl-lactate.pdf

Generated by Cheméo on 2024-04-09 23:25:11.746169634 +0000 UTC m=+14994360.666746949.

Cheméo (https://www.chemeo.com) is the biggest free database of chemical and physical data for the process industry.