

# methyl 3-hydroxy-2-methylbutanoate

<b>Other names:</b>	methyl 3-hydroxy-2-methylbutyrate
<b>Inchi:</b>	InChI=1S/C6H12O3/c1-4(5(2)7)6(8)9-3/h4-5,7H,1-3H3
<b>InchiKey:</b>	FFJMPYODEQVBEX-UHFFFAOYSA-N
<b>Formula:</b>	C6H12O3
<b>SMILES:</b>	<chem>COC(=O)C(C)C(C)O</chem>
<b>Mol. weight [g/mol]:</b>	132.16

## Physical Properties

Property code	Value	Unit	Source
gf	-375.98	kJ/mol	Joback Method
hf	-574.76	kJ/mol	Joback Method
hfus	11.12	kJ/mol	Joback Method
hvap	54.01	kJ/mol	Joback Method
log10ws	-0.33		Crippen Method
logp	0.176		Crippen Method
mcvol	108.710	ml/mol	McGowan Method
pc	3718.02	kPa	Joback Method
rinpol	930.00		NIST Webbook
rinpol	901.00		NIST Webbook
rinpol	930.00		NIST Webbook
rinpol	901.00		NIST Webbook
ripol	1518.00		NIST Webbook
ripol	1509.00		NIST Webbook
ripol	1519.00		NIST Webbook
ripol	1534.00		NIST Webbook
ripol	1523.00		NIST Webbook
ripol	1509.00		NIST Webbook
ripol	1519.00		NIST Webbook
tb	504.27	K	Joback Method
tc	682.14	K	Joback Method
tf	260.36	K	Joback Method
vc	0.403	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	246.04	J/molxK	504.27	Joback Method
cpg	255.38	J/molxK	533.92	Joback Method
cpg	264.36	J/molxK	563.56	Joback Method
cpg	272.99	J/molxK	593.21	Joback Method
cpg	281.27	J/molxK	622.85	Joback Method
cpg	289.19	J/molxK	652.50	Joback Method
cpg	296.77	J/molxK	682.14	Joback Method
dvisc	0.0398736	Paxs	260.36	Joback Method
dvisc	0.0082434	Paxs	301.01	Joback Method
dvisc	0.0024799	Paxs	341.66	Joback Method
dvisc	0.0009632	Paxs	382.31	Joback Method
dvisc	0.0004487	Paxs	422.97	Joback Method
dvisc	0.0002390	Paxs	463.62	Joback Method
dvisc	0.0001409	Paxs	504.27	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R319474&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R319474&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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