

# 1-(4-Hydroxy-3-methoxyphenyl)octane-3,5-diyl

InChI:	InChI=1S/C19H28O6/c1-5-6-16(24-13(2)20)12-17(25-14(3)21)9-7-15-8-10-18(22)19(11-
diacetate	
InChIKey:	AUBPDZJRJKZQEX-UHFFFAOYSA-N
Formula:	C19H28O6
SMILES:	CCCC(CC(CCc1ccc(O)c(OC)c1)OC(C)=O)OC(C)=O
Mol. weight [g/mol]:	352.42
CAS:	53254-50-5

## Physical Properties

Property code	Value	Unit	Source
gf	-520.46	kJ/mol	Joback Method
hf	-1020.12	kJ/mol	Joback Method
hfus	44.12	kJ/mol	Joback Method
hvap	93.79	kJ/mol	Joback Method
log10ws	-4.08		Crippen Method
logp	3.387		Crippen Method
mvol	281.430	ml/mol	McGowan Method
pc	1618.07	kPa	Joback Method
rinpol	2357.30		NIST Webbook
tb	920.52	K	Joback Method
tc	1136.08	K	Joback Method
tf	591.10	K	Joback Method
vc	1.012	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	904.46	J/molxK	920.52	Joback Method
cpg	918.89	J/molxK	956.45	Joback Method
cpg	932.30	J/molxK	992.37	Joback Method
cpg	944.75	J/molxK	1028.30	Joback Method
cpg	956.27	J/molxK	1064.23	Joback Method
cpg	966.92	J/molxK	1100.15	Joback Method
cpg	976.74	J/molxK	1136.08	Joback Method
dvisc	0.0000444	Paxs	591.10	Joback Method

dvisc	0.0000200	Paxs	646.00	Joback Method
dvisc	0.0000103	Paxs	700.91	Joback Method
dvisc	0.0000058	Paxs	755.81	Joback Method
dvisc	0.0000035	Paxs	810.71	Joback Method
dvisc	0.0000023	Paxs	865.62	Joback Method
dvisc	0.0000016	Paxs	920.52	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C53254505&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C53254505&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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