

# Succinic acid, hept-2-yl 2-methoxy-5-methylphenyl ester

<b>Inchi:</b>	InChI=1S/C19H28O5/c1-5-6-7-8-15(3)23-18(20)11-12-19(21)24-17-13-14(2)9-10-16(17)2
<b>InchiKey:</b>	UOYBWQDIOPIYRO-UHFFFAOYSA-N
<b>Formula:</b>	C19H28O5
<b>SMILES:</b>	CCCCC(C)OC(=O)CCC(=O)Oc1cc(C)ccc1OC
<b>Mol. weight [g/mol]:</b>	336.42

## Physical Properties

Property code	Value	Unit	Source
gf	-373.03	kJ/mol	Joback Method
hf	-849.00	kJ/mol	Joback Method
hfus	41.47	kJ/mol	Joback Method
hvap	81.82	kJ/mol	Joback Method
log10ws	-5.12		Crippen Method
logp	4.201		Crippen Method
mvol	275.560	ml/mol	McGowan Method
pc	1415.44	kPa	Joback Method
rinpol	2350.00		NIST Webbook
rinpol	2350.00		NIST Webbook
tb	845.32	K	Joback Method
tc	1048.37	K	Joback Method
tf	506.90	K	Joback Method
vc	1.052	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	847.95	J/molxK	845.32	Joback Method
cpg	863.44	J/molxK	879.16	Joback Method
cpg	877.73	J/molxK	913.00	Joback Method
cpg	890.84	J/molxK	946.84	Joback Method
cpg	902.75	J/molxK	980.68	Joback Method
cpg	913.49	J/molxK	1014.53	Joback Method
cpg	923.04	J/molxK	1048.37	Joback Method
dvisc	0.0004436	Paxs	506.90	Joback Method

dvisc	0.0002488	Paxs	563.30	Joback Method
dvisc	0.0001551	Paxs	619.71	Joback Method
dvisc	0.0001046	Paxs	676.11	Joback Method
dvisc	0.0000749	Paxs	732.51	Joback Method
dvisc	0.0000563	Paxs	788.92	Joback Method
dvisc	0.0000440	Paxs	845.32	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390959&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390959&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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