

# 1H-Pyrazole, 4,5-dihydro-3,5,5-trimethyl-

<b>Other names:</b>	2-Pyrazoline, 3,5,5-trimethyl- 3,5,5-Trimethyl-2-pyrazoline 3,5,5-Trimethylpyrazoline 3,5,5-Trimethyl-«DELTA»[2]-pyrazoline
<b>Inchi:</b>	InChI=1S/C6H12N2/c1-5-4-6(2,3)8-7-5/h8H,4H2,1-3H3
<b>InchiKey:</b>	MEVTUZFUVAZEPEY-UHFFFAOYSA-N
<b>Formula:</b>	C6H12N2
<b>SMILES:</b>	CC1=NNC(C)(C)C1
<b>Mol. weight [g/mol]:</b>	112.17
<b>CAS:</b>	3975-85-7

## Physical Properties

Property code	Value	Unit	Source
gf	255.52	kJ/mol	Joback Method
hf	63.64	kJ/mol	Joback Method
hfus	14.49	kJ/mol	Joback Method
hvap	41.98	kJ/mol	Joback Method
log10ws	-1.68		Crippen Method
logp	1.134		Crippen Method
mcvol	100.200	ml/mol	McGowan Method
pc	4316.89	kPa	Joback Method
rinpol	910.00		NIST Webbook
rinpol	910.00		NIST Webbook
tb	458.59	K	Joback Method
tc	690.08	K	Joback Method
tf	382.03	K	Joback Method
vc	0.383	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	216.75	J/mol×K	458.59	Joback Method
cpg	231.50	J/mol×K	497.17	Joback Method
cpg	245.27	J/mol×K	535.75	Joback Method

cpg	258.18	J/mol×K	574.33	Joback Method
cpg	270.32	J/mol×K	612.91	Joback Method
cpg	281.78	J/mol×K	651.49	Joback Method
cpg	292.66	J/mol×K	690.08	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3975857&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3975857&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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