

# 1-carbomethoxy-3-methylcyclohex-3-ene

<b>Inchi:</b>	InChI=1S/C9H14O2/c1-7-4-3-5-8(6-7)9(10)11-2/h4,8H,3,5-6H2,1-2H3
<b>InchiKey:</b>	ALWNXIRSKYWVFP-UHFFFAOYSA-N
<b>Formula:</b>	C9H14O2
<b>SMILES:</b>	COC(=O)C1CCC=C(C)C1
<b>Mol. weight [g/mol]:</b>	154.21

## Physical Properties

Property code	Value	Unit	Source
gf	-164.24	kJ/mol	Joback Method
hf	-373.26	kJ/mol	Joback Method
hfus	14.52	kJ/mol	Joback Method
hvap	46.17	kJ/mol	Joback Method
log10ws	-1.96		Crippen Method
logp	1.906		Crippen Method
mcvol	129.950	ml/mol	McGowan Method
pc	3052.41	kPa	Joback Method
ripol	1549.30		NIST Webbook
tb	505.30	K	Joback Method
tc	717.03	K	Joback Method
tf	284.01	K	Joback Method
vc	0.482	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	291.84	J/molxK	505.30	Joback Method
cpg	307.31	J/molxK	540.59	Joback Method
cpg	322.04	J/molxK	575.88	Joback Method
cpg	336.01	J/molxK	611.16	Joback Method
cpg	349.24	J/molxK	646.45	Joback Method
cpg	361.73	J/molxK	681.74	Joback Method
cpg	373.48	J/molxK	717.03	Joback Method
dvisc	0.0025360	Paxs	284.01	Joback Method
dvisc	0.0013801	Paxs	320.89	Joback Method

dvisc	0.0008515	Paxs	357.77	Joback Method
dvisc	0.0005749	Paxs	394.65	Joback Method
dvisc	0.0004152	Paxs	431.54	Joback Method
dvisc	0.0003156	Paxs	468.42	Joback Method
dvisc	0.0002497	Paxs	505.30	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R388282&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R388282&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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