

# Cyclopentanemethanol, 1-amino-

<b>Other names:</b>	1-Aminocyclopentanemethanol Amino-1 methylol-1 cyclopentane 1-Amino-1-cyclopentanemethanol
<b>Inchi:</b>	InChI=1S/C6H13NO/c7-6(5-8)3-1-2-4-6/h8H,1-5,7H2
<b>InchiKey:</b>	PDNZJLMPXLQDPL-UHFFFAOYSA-N
<b>Formula:</b>	C6H13NO
<b>SMILES:</b>	NC1(CO)CCCC1
<b>Mol. weight [g/mol]:</b>	115.17
<b>CAS:</b>	10316-79-7

## Physical Properties

Property code	Value	Unit	Source
gf	-39.67	kJ/mol	Joback Method
hf	-209.89	kJ/mol	Joback Method
hfus	8.22	kJ/mol	Joback Method
hvap	55.38	kJ/mol	Joback Method
log10ws	-1.04		Crippen Method
logp	0.250		Crippen Method
mcvol	100.390	ml/mol	McGowan Method
pc	4959.33	kPa	Joback Method
tb	516.91	K	Joback Method
tc	724.20	K	Joback Method
tf	431.50 ± 0.50	K	NIST Webbook
vc	0.358	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	244.76	J/mol×K	516.91	Joback Method
cpg	256.66	J/mol×K	551.46	Joback Method
cpg	267.74	J/mol×K	586.01	Joback Method
cpg	278.10	J/mol×K	620.56	Joback Method
cpg	287.84	J/mol×K	655.10	Joback Method
cpg	297.08	J/mol×K	689.65	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	360.20	K	1.30	NIST Webbook
tbrp	341.50 ± 0.50	K	0.10	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10316797&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10316797&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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