

# Propionitrile, 3-trimethylhydrazino-

<b>Other names:</b>	1,1-Dimethyl-2-(2-cyanoprop-2-yl)hydrazine
<b>Inchi:</b>	InChI=1S/C6H13N3/c1-6(2,5-7)8-9(3)4/h8H,1-4H3
<b>InchiKey:</b>	AJMSTEDXUWYXGY-UHFFFAOYSA-N
<b>Formula:</b>	C6H13N3
<b>SMILES:</b>	CN(C)NC(C)(C)C#N
<b>Mol. weight [g/mol]:</b>	127.19
<b>CAS:</b>	97812-31-2

## Physical Properties

Property code	Value	Unit	Source
gf	335.83	kJ/mol	Joback Method
hf	109.96	kJ/mol	Joback Method
hfus	13.51	kJ/mol	Joback Method
hvap	46.61	kJ/mol	Joback Method
ie	9.00	eV	NIST Webbook
log10ws	-1.06		Crippen Method
logp	0.355		Crippen Method
mcvol	116.740	ml/mol	McGowan Method
pc	3124.49	kPa	Joback Method
tb	498.14	K	Joback Method
tc	699.42	K	Joback Method
tf	309.92	K	Joback Method
vc	0.440	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	265.24	J/molxK	498.14	Joback Method
cpg	276.99	J/molxK	531.69	Joback Method
cpg	288.00	J/molxK	565.23	Joback Method
cpg	298.31	J/molxK	598.78	Joback Method
cpg	307.96	J/molxK	632.33	Joback Method
cpg	317.00	J/molxK	665.87	Joback Method
cpg	325.45	J/molxK	699.42	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C97812312&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C97812312&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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