

# Phenol, 4-(ethoxymethyl)-

<b>Other names:</b>	4-Ethoxymethylphenol «alpha»-ethoxy-p-cresol
<b>Inchi:</b>	InChI=1S/C9H12O2/c1-2-11-7-8-3-5-9(10)6-4-8/h3-6,10H,2,7H2,1H3
<b>InchiKey:</b>	UWQZVUQKBWZNLN-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O2
<b>SMILES:</b>	CCOCc1ccc(O)cc1
<b>Mol. weight [g/mol]:</b>	152.19
<b>CAS:</b>	57726-26-8

## Physical Properties

Property code	Value	Unit	Source
gf	-122.31	kJ/mol	Joback Method
hf	-302.09	kJ/mol	Joback Method
hfus	20.08	kJ/mol	Joback Method
hvap	53.33	kJ/mol	Joback Method
log10ws	-1.82		Crippen Method
logp	1.929		Crippen Method
mcvol	125.650	ml/mol	McGowan Method
pc	3815.10	kPa	Joback Method
rinpol	1378.70		NIST Webbook
rinpol	1378.70		NIST Webbook
tb	535.04	K	Joback Method
tc	754.68	K	Joback Method
tf	351.56	K	Joback Method
vc	0.415	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	290.27	J/molxK	535.04	Joback Method
cpg	345.78	J/molxK	718.08	Joback Method
cpg	336.05	J/molxK	681.47	Joback Method
cpg	325.69	J/molxK	644.86	Joback Method
cpg	314.65	J/molxK	608.25	Joback Method

cpg	302.86	J/molxK	571.65	Joback Method
cpg	354.94	J/molxK	754.68	Joback Method
dvisc	0.0000540	Paxs	535.04	Joback Method
dvisc	0.0000833	Paxs	504.46	Joback Method
dvisc	0.0001359	Paxs	473.88	Joback Method
dvisc	0.0002372	Paxs	443.30	Joback Method
dvisc	0.0004496	Paxs	412.72	Joback Method
dvisc	0.0009440	Paxs	382.14	Joback Method
dvisc	0.0022554	Paxs	351.56	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C57726268&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C57726268&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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