

# Phthalic acid, undecyl 2-trifluoromethylbenzyl ester

<b>Other names:</b>	Phthalic acid, 2-trifluorobenzyl undecyl ester
<b>Inchi:</b>	InChI=1S/C27H33F3O4/c1-2-3-4-5-6-7-8-9-14-19-33-25(31)22-16-11-12-17-23(22)26(32)
<b>InchiKey:</b>	QIZDENQBWLXPIO-UHFFFAOYSA-N
<b>Formula:</b>	C27H33F3O4
<b>SMILES:</b>	CCCCCCCCCCCCOC(=O)c1ccccc1C(=O)OCc1ccccc1C(F)(F)F
<b>Mol. weight [g/mol]:</b>	478.54

## Physical Properties

Property code	Value	Unit	Source
gf	-667.41	kJ/mol	Joback Method
hf	-1237.17	kJ/mol	Joback Method
hfus	60.39	kJ/mol	Joback Method
hvap	96.14	kJ/mol	Joback Method
log10ws	-9.36		Crippen Method
logp	7.750		Crippen Method
mcvol	363.960	ml/mol	McGowan Method
pc	975.34	kPa	Joback Method
rinpol	3009.00		NIST Webbook
rinpol	3009.00		NIST Webbook
tb	1027.64	K	Joback Method
tc	1258.34	K	Joback Method
tf	620.44	K	Joback Method
vc	1.423	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1218.03	J/molxK	1027.64	Joback Method
cpg	1232.31	J/molxK	1066.09	Joback Method
cpg	1245.21	J/molxK	1104.54	Joback Method
cpg	1256.84	J/molxK	1142.99	Joback Method
cpg	1267.28	J/molxK	1181.44	Joback Method
cpg	1276.62	J/molxK	1219.89	Joback Method
cpg	1284.97	J/molxK	1258.34	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U377826&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U377826&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/92-177-4/Phthalic-acid-undecyl-2-trifluoromethylbenzyl-ester.pdf>

Generated by Cheméo on 2024-04-27 10:19:56.724920216 +0000 UTC m=+16502445.645497531.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.