

# (3-Iodophenyl) methanol, ethyl ether

<b>Inchi:</b>	InChI=1S/C9H11IO/c1-2-11-7-8-4-3-5-9(10)6-8/h3-6H,2,7H2,1H3
<b>InchiKey:</b>	OSYGCJIBBWLFQ-UHFFFAOYSA-N
<b>Formula:</b>	C9H11IO
<b>SMILES:</b>	CCOCc1cccc(I)c1
<b>Mol. weight [g/mol]:</b>	262.09

## Physical Properties

Property code	Value	Unit	Source
gf	80.80	kJ/mol	Joback Method
hf	-59.38	kJ/mol	Joback Method
hfus	18.31	kJ/mol	Joback Method
hvap	50.35	kJ/mol	Joback Method
log10ws	-3.42		Crippen Method
logp	2.828		Crippen Method
mvol	145.600	ml/mol	McGowan Method
pc	3069.34	kPa	Joback Method
rinpol	1458.00		NIST Webbook
rinpol	1458.00		NIST Webbook
tb	552.54	K	Joback Method
tc	792.32	K	Joback Method
tf	310.42	K	Joback Method
vc	0.537	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	288.52	J/molxK	552.54	Joback Method
cpg	301.23	J/molxK	592.50	Joback Method
cpg	313.12	J/molxK	632.47	Joback Method
cpg	324.23	J/molxK	672.43	Joback Method
cpg	334.59	J/molxK	712.40	Joback Method
cpg	344.23	J/molxK	752.36	Joback Method
cpg	353.17	J/molxK	792.32	Joback Method
dvisc	0.0021690	Paxs	310.42	Joback Method

dvisc	0.0011910	Paxs	350.77	Joback Method
dvisc	0.0007401	Paxs	391.13	Joback Method
dvisc	0.0005027	Paxs	431.48	Joback Method
dvisc	0.0003648	Paxs	471.83	Joback Method
dvisc	0.0002785	Paxs	512.19	Joback Method
dvisc	0.0002211	Paxs	552.54	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U374570&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U374570&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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