

# Benzeneacetic acid, (4-methoxyphenyl)methyl ester

**Other names:**

Acetic acid, phenyl-, p-methoxybenzyl ester

p-Anisyl phenylacetate

p-Methoxybenzyl phenylacetate

Phenylacetic acid, p-methoxybenzyl ester

4-methoxybenzyl phenylacetate

**Inchi:** InChI=1S/C16H16O3/c1-18-15-9-7-14(8-10-15)12-19-16(17)11-13-5-3-2-4-6-13/h2-10H,1**InchiKey:** VCYWCSZLXMMLLE-UHFFFAOYSA-N**Formula:** C16H16O3**SMILES:** COc1ccc(COC(=O)Cc2ccccc2)cc1**Mol. weight [g/mol]:** 256.30**CAS:** 102-17-0

## Physical Properties

Property code	Value	Unit	Source
gf	-39.89	kJ/mol	Joback Method
hf	-289.00	kJ/mol	Joback Method
hfus	28.86	kJ/mol	Joback Method
hvap	67.99	kJ/mol	Joback Method
log10ws	-3.79		Crippen Method
logp	2.981		Crippen Method
mcvol	202.090	ml/mol	McGowan Method
pc	2315.84	kPa	Joback Method
tb	722.53	K	Joback Method
tc	953.08	K	Joback Method
tf	429.83	K	Joback Method
vc	0.757	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	544.43	J/molxK	722.53	Joback Method
cpg	559.93	J/molxK	760.96	Joback Method
cpg	574.23	J/molxK	799.38	Joback Method
cpg	587.37	J/molxK	837.81	Joback Method

cpg	599.38	J/molxK	876.23	Joback Method
cpg	610.29	J/molxK	914.66	Joback Method
cpg	620.12	J/molxK	953.08	Joback Method
dvisc	0.0008483	Paxs	429.83	Joback Method
dvisc	0.0004890	Paxs	478.61	Joback Method
dvisc	0.0003121	Paxs	527.40	Joback Method
dvisc	0.0002150	Paxs	576.18	Joback Method
dvisc	0.0001569	Paxs	624.96	Joback Method
dvisc	0.0001199	Paxs	673.75	Joback Method
dvisc	0.0000950	Paxs	722.53	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C102170&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C102170&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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