

# Propanoic acid, 2-phenoxy-

<b>Other names:</b>	Propionic acid, 2-phenoxy- «alpha»-Methylphenoxyacetic acid «alpha»-Phenoxypropionic acid 2-Phenoxypropionic acid dl-2-Phenoxypropionic acid Acide phenoxy-2-propionique
<b>Inchi:</b>	InChI=1S/C9H10O3/c1-7(9(10)11)12-8-5-3-2-4-6-8/h2-7H,1H3,(H,10,11)
<b>InchiKey:</b>	SXERGJJQSKIUIC-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O3
<b>SMILES:</b>	CC(Oc1ccccc1)C(=O)O
<b>Mol. weight [g/mol]:</b>	166.17
<b>CAS:</b>	940-31-8

## Physical Properties

Property code	Value	Unit	Source
gf	-235.87	kJ/mol	Joback Method
hf	-394.87	kJ/mol	Joback Method
hfus	16.46	kJ/mol	Joback Method
hvap	63.35	kJ/mol	Joback Method
log10ws	-1.64		Crippen Method
logp	1.538		Crippen Method
mvol	127.220	ml/mol	McGowan Method
pc	3881.95	kPa	Joback Method
rinpol	1372.00		NIST Webbook
rinpol	1372.00		NIST Webbook
tb	538.20	K	NIST Webbook
tc	805.31	K	Joback Method
tf	335.59	K	Joback Method
vc	0.469	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	351.39	J/mol×K	771.10	Joback Method

cpg	343.24	J/mol×K	736.89	Joback Method
cpg	334.50	J/mol×K	702.67	Joback Method
cpg	325.16	J/mol×K	668.46	Joback Method
cpg	315.18	J/mol×K	634.24	Joback Method
cpg	304.57	J/mol×K	600.03	Joback Method
cpg	358.96	J/mol×K	805.31	Joback Method
dvisc	0.0059290	Paxs	335.59	Joback Method
dvisc	0.0000761	Paxs	600.03	Joback Method
dvisc	0.0001179	Paxs	555.96	Joback Method
dvisc	0.0001971	Paxs	511.88	Joback Method
dvisc	0.0003629	Paxs	467.81	Joback Method
dvisc	0.0007586	Paxs	423.74	Joback Method
dvisc	0.0018822	Paxs	379.66	Joback Method
hfust	33.05	kJ/mol	388.00	NIST Webbook
hfust	33.05	kJ/mol	388.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C940318&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C940318&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/92-220-5/Propanoic-acid-2-phenoxy.pdf>

Generated by Cheméo on 2024-04-25 19:33:37.730438836 +0000 UTC m=+16362866.651016152.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.