

# Cyclopentanecarboxylic acid, 2-naphthyl ester

<b>Inchi:</b>	InChI=1S/C16H16O2/c17-16(13-6-2-3-7-13)18-15-10-9-12-5-1-4-8-14(12)11-15/h1,4-5,8
<b>InchiKey:</b>	SKMMIZABGFBLRQ-UHFFFAOYSA-N
<b>Formula:</b>	C16H16O2
<b>SMILES:</b>	O=C(Oc1ccc2ccccc2c1)C1CCCC1
<b>Mol. weight [g/mol]:</b>	240.30

## Physical Properties

Property code	Value	Unit	Source
gf	95.90	kJ/mol	Joback Method
hf	-141.76	kJ/mol	Joback Method
hfus	24.59	kJ/mol	Joback Method
hvap	65.20	kJ/mol	Joback Method
log10ws	-4.92		Crippen Method
logp	3.935		Crippen Method
mvol	189.660	ml/mol	McGowan Method
pc	2584.59	kPa	Joback Method
rinpol	2058.00		NIST Webbook
rinpol	2058.00		NIST Webbook
tb	707.69	K	Joback Method
tc	954.96	K	Joback Method
tf	424.78	K	Joback Method
vc	0.711	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	526.62	J/molxK	707.69	Joback Method
cpg	544.01	J/molxK	748.90	Joback Method
cpg	560.00	J/molxK	790.11	Joback Method
cpg	574.68	J/molxK	831.32	Joback Method
cpg	588.15	J/molxK	872.53	Joback Method
cpg	600.51	J/molxK	913.74	Joback Method
cpg	611.87	J/molxK	954.96	Joback Method
dvisc	0.0016903	Paxs	424.78	Joback Method

dvisc	0.0011119	Paxs	471.93	Joback Method
dvisc	0.0007892	Paxs	519.08	Joback Method
dvisc	0.0005931	Paxs	566.24	Joback Method
dvisc	0.0004657	Paxs	613.39	Joback Method
dvisc	0.0003786	Paxs	660.54	Joback Method
dvisc	0.0003163	Paxs	707.69	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307581&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307581&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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