

# Propanoic acid, 2-phenoxy-, (.+/-.)-

|                      |                                                                    |
|----------------------|--------------------------------------------------------------------|
| Inchi:               | InChI=1S/C9H10O3/c1-7(9(10)11)12-8-5-3-2-4-6-8/h2-7H,1H3,(H,10,11) |
| InchiKey:            | SXERGJJQSKIUIC-UHFFFAOYSA-N                                        |
| Formula:             | C9H10O3                                                            |
| SMILES:              | CC(Oc1ccccc1)C(=O)O                                                |
| Mol. weight [g/mol]: | 166.17                                                             |
| CAS:                 | 1912-21-6                                                          |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -235.87 | kJ/mol  | Joback Method  |
| hf            | -394.87 | kJ/mol  | Joback Method  |
| hfus          | 16.46   | kJ/mol  | Joback Method  |
| hvap          | 63.35   | kJ/mol  | Joback Method  |
| log10ws       | -1.64   |         | Crippen Method |
| logp          | 1.538   |         | Crippen Method |
| mcvol         | 127.220 | ml/mol  | McGowan Method |
| pc            | 3881.95 | kPa     | Joback Method  |
| tb            | 538.70  | K       | NIST Webbook   |
| tc            | 805.31  | K       | Joback Method  |
| tf            | 335.59  | K       | Joback Method  |
| vc            | 0.469   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 304.57    | J/mol×K | 600.03          | Joback Method |
| cpg           | 351.39    | J/mol×K | 771.10          | Joback Method |
| cpg           | 343.24    | J/mol×K | 736.89          | Joback Method |
| cpg           | 334.50    | J/mol×K | 702.67          | Joback Method |
| cpg           | 325.16    | J/mol×K | 668.46          | Joback Method |
| cpg           | 315.18    | J/mol×K | 634.24          | Joback Method |
| cpg           | 358.96    | J/mol×K | 805.31          | Joback Method |
| dvisc         | 0.0000761 | Pa×s    | 600.03          | Joback Method |
| dvisc         | 0.0001179 | Pa×s    | 555.96          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0001971 | Pa×s | 511.88 | Joback Method |
| dvisc | 0.0003629 | Pa×s | 467.81 | Joback Method |
| dvisc | 0.0007586 | Pa×s | 423.74 | Joback Method |
| dvisc | 0.0018822 | Pa×s | 379.66 | Joback Method |
| dvisc | 0.0059290 | Pa×s | 335.59 | Joback Method |

## Pressure Dependent Properties

| Property code | Value  | Unit | Pressure [kPa] | Source       |
|---------------|--------|------|----------------|--------------|
| tbrp          | 378.70 | K    | 0.70           | NIST Webbook |

## Sources

|                 |                                                                                                                                             |
|-----------------|---------------------------------------------------------------------------------------------------------------------------------------------|
| NIST Webbook:   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1912216&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1912216&amp;Units=SI</a> |
| Crippen Method: | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                   |
| Crippen Method: | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                           |
| Joback Method:  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                       |
| McGowan Method: | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                       |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tbrp:</b>    | Boiling point at reduced pressure               |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |

**vc:** Critical Volume

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