

# 1-hydroxy-3-phenyl-2-propanone

<b>Inchi:</b>	InChI=1S/C9H10O2/c10-7-9(11)6-8-4-2-1-3-5-8/h1-5,10H,6-7H2
<b>InchiKey:</b>	QLCZRWKIQPLYFS-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O2
<b>SMILES:</b>	O=C(CO)Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	150.17

## Physical Properties

Property code	Value	Unit	Source
gf	-128.43	kJ/mol	Joback Method
hf	-257.37	kJ/mol	Joback Method
hfus	18.79	kJ/mol	Joback Method
hvap	61.33	kJ/mol	Joback Method
log10ws	-1.24		Crippen Method
logp	0.790		Crippen Method
mcvol	121.350	ml/mol	McGowan Method
pc	3925.85	kPa	Joback Method
ripol	2303.00		NIST Webbook
ripol	2303.00		NIST Webbook
tb	578.05	K	Joback Method
tc	781.24	K	Joback Method
tf	328.36	K	Joback Method
vc	0.457	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	281.36	J/molxK	578.05	Joback Method
cpg	291.87	J/molxK	611.92	Joback Method
cpg	301.73	J/molxK	645.78	Joback Method
cpg	310.96	J/molxK	679.65	Joback Method
cpg	319.60	J/molxK	713.51	Joback Method
cpg	327.67	J/molxK	747.38	Joback Method
cpg	335.20	J/molxK	781.24	Joback Method
dvisc	0.0068844	Paxs	328.36	Joback Method

dvisc	0.0023345	Paxs	369.98	Joback Method
dvisc	0.0009851	Paxs	411.59	Joback Method
dvisc	0.0004871	Paxs	453.20	Joback Method
dvisc	0.0002711	Paxs	494.82	Joback Method
dvisc	0.0001653	Paxs	536.43	Joback Method
dvisc	0.0001082	Paxs	578.05	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R389725&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R389725&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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