

# Pentanedioic acid, monomethyl ester

<b>Other names:</b>	Glutaric acid, monomethyl ester Methyl hydrogen glutarate Monomethyl glutarate Methyl glutarate,mono Monomethyl ester of glutaric acid
<b>Inchi:</b>	InChI=1S/C6H10O4/c1-10-6(9)4-2-3-5(7)8/h2-4H2,1H3,(H,7,8)
<b>InchiKey:</b>	IBMRTYCHDPMBFN-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O4
<b>SMILES:</b>	COC(=O)CCCC(=O)O
<b>Mol. weight [g/mol]:</b>	146.14
<b>CAS:</b>	1501-27-5

## Physical Properties

Property code	Value	Unit	Source
gf	-500.02	kJ/mol	Joback Method
hf	-676.78	kJ/mol	Joback Method
hfus	19.77	kJ/mol	Joback Method
hvap	61.53	kJ/mol	Joback Method
log10ws	-0.30		Crippen Method
logp	0.414		Crippen Method
mcvol	110.280	ml/mol	McGowan Method
pc	3940.66	kPa	Joback Method
tb	559.02	K	Joback Method
tc	738.59	K	Joback Method
tf	340.29	K	Joback Method
vc	0.420	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.90	J/molxK	559.02	Joback Method
cpg	266.10	J/molxK	588.95	Joback Method
cpg	273.95	J/molxK	618.88	Joback Method
cpg	281.47	J/molxK	648.80	Joback Method

cpg	288.64	J/molxK	678.73	Joback Method
cpg	295.48	J/molxK	708.66	Joback Method
cpg	301.97	J/molxK	738.59	Joback Method
dvisc	0.0052745	Paxs	340.29	Joback Method
dvisc	0.0021175	Paxs	376.75	Joback Method
dvisc	0.0009987	Paxs	413.20	Joback Method
dvisc	0.0005320	Paxs	449.65	Joback Method
dvisc	0.0003115	Paxs	486.11	Joback Method
dvisc	0.0001965	Paxs	522.56	Joback Method
dvisc	0.0001317	Paxs	559.02	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	423.70	K	1.30	NIST Webbook
tbrp	434.50 ± 3.50	K	3.10	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1501275&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1501275&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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