

# 2-(Chloromethyl)tetrahydropyran

<b>Other names:</b>	2-(Chloromethyl)tetrahydro-2H-pyran
<b>Inchi:</b>	InChI=1S/C6H11ClO/c7-5-6-3-1-2-4-8-6/h6H,1-5H2
<b>InchiKey:</b>	PPYKTTGONDVGPX-UHFFFAOYSA-N
<b>Formula:</b>	C6H11ClO
<b>SMILES:</b>	CICC1CCCCO1
<b>Mol. weight [g/mol]:</b>	134.60
<b>CAS:</b>	18420-41-2

## Physical Properties

Property code	Value	Unit	Source
gf	-73.96	kJ/mol	Joback Method
hf	-260.59	kJ/mol	Joback Method
hfus	15.31	kJ/mol	Joback Method
hvap	38.27	kJ/mol	Joback Method
log10ws	-1.58		Crippen Method
logp	1.794		Crippen Method
mcvol	102.650	ml/mol	McGowan Method
pc	3740.80	kPa	Joback Method
tb	420.61	K	Joback Method
tc	633.59	K	Joback Method
tf	221.25	K	Joback Method
vc	0.374	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	192.54	J/mol×K	420.61	Joback Method
cpg	206.68	J/mol×K	456.11	Joback Method
cpg	220.07	J/mol×K	491.60	Joback Method
cpg	232.74	J/mol×K	527.10	Joback Method
cpg	244.70	J/mol×K	562.60	Joback Method
cpg	255.98	J/mol×K	598.10	Joback Method
cpg	266.59	J/mol×K	633.59	Joback Method
dvisc	0.0075317	Paxs	221.25	Joback Method

dvisc	0.0033009	Paxs	254.48	Joback Method
dvisc	0.0017503	Paxs	287.70	Joback Method
dvisc	0.0010584	Paxs	320.93	Joback Method
dvisc	0.0007034	Paxs	354.16	Joback Method
dvisc	0.0005014	Paxs	387.38	Joback Method
dvisc	0.0003770	Paxs	420.61	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	326.70	K	1.60	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18420412&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18420412&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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