

# 2,2,4-Trimethyl-1,3-pentanediol diisobutyrate

<b>Other names:</b>	Propanoic acid, 2-methyl-, 2,2-dimethyl-1-(1-methylethyl)-1,3-propanediyl ester 1,3-Pentanediol, 2,2,4-trimethyl-, diisobutyrate Kodaflex txib Isobutyric acid, 1-isopropyl-2,2-dimethyltrimethylene ester 2,2,4-Trimethylpentanediol-1,3-diisobutyrate Txib 1-isopropyl-2,2-dimethyltrimethylene diisobutyrate
<b>Inchi:</b>	InChI=1S/C16H30O4/c1-10(2)13(20-15(18)12(5)6)16(7,8)9-19-14(17)11(3)4/h10-13H,9H
<b>InchiKey:</b>	OMVSWZDEEGIJJI-UHFFFAOYSA-N
<b>Formula:</b>	C16H30O4
<b>SMILES:</b>	CC(C)C(=O)OCC(C)(C)C(OC(=O)C(C)C)C(C)C
<b>Mol. weight [g/mol]:</b>	286.41
<b>CAS:</b>	6846-50-0

## Physical Properties

Property code	Value	Unit	Source
gf	-390.92	kJ/mol	Joback Method
hf	-893.04	kJ/mol	Joback Method
hfus	21.26	kJ/mol	Joback Method
hvap	66.67	kJ/mol	Joback Method
log10ws	-3.39		Crippen Method
logp	3.436		Crippen Method
mcvol	251.180	ml/mol	McGowan Method
pc	1466.85	kPa	Joback Method
rinpol	1591.00		NIST Webbook
rinpol	1587.50		NIST Webbook
rinpol	1587.50		NIST Webbook
tb	713.07	K	Joback Method
tc	904.59	K	Joback Method
tf	356.82	K	Joback Method
vc	0.945	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	737.61	J/mol×K	713.07	Joback Method
cpg	816.90	J/mol×K	872.67	Joback Method
cpg	802.99	J/mol×K	840.75	Joback Method
cpg	788.13	J/mol×K	808.83	Joback Method
cpg	772.29	J/mol×K	776.91	Joback Method
cpg	755.46	J/mol×K	744.99	Joback Method
cpg	829.89	J/mol×K	904.59	Joback Method
dvisc	0.0000565	Paxs	713.07	Joback Method
dvisc	0.0000822	Paxs	653.70	Joback Method
dvisc	0.0001289	Paxs	594.32	Joback Method
dvisc	0.0002234	Paxs	534.95	Joback Method
dvisc	0.0004442	Paxs	475.57	Joback Method
dvisc	0.0010748	Paxs	416.20	Joback Method
dvisc	0.0034893	Paxs	356.82	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6846500&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6846500&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/92-437-5/2-2-4-Trimethyl-1-3-pentanediol-diisobutyrate.pdf>

Generated by Cheméo on 2024-04-27 04:49:30.998868828 +0000 UTC m=+16482619.919446143.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.