

# 2-Propenoic acid, 2-methyl-, ethenyl ester

<b>Other names:</b>	Methacrylic acid, vinyl ester Vinyl methacrylate 2-Methyl-2-propenoic acid, ethenyl ester 1-ethenyl-2-methyl-2-propenoate
<b>Inchi:</b>	InChI=1S/C6H8O2/c1-4-8-6(7)5(2)3/h4H,1-2H2,3H3
<b>InchiKey:</b>	FFYWKOUKJFCBAM-UHFFFAOYSA-N
<b>Formula:</b>	C6H8O2
<b>SMILES:</b>	<chem>C=COC(=O)C(=C)C</chem>
<b>Mol. weight [g/mol]:</b>	112.13
<b>CAS:</b>	4245-37-8

## Physical Properties

Property code	Value	Unit	Source
gf	-67.15	kJ/mol	Joback Method
hf	-170.90	kJ/mol	Joback Method
hfus	10.21	kJ/mol	Joback Method
hvap	36.85	kJ/mol	Joback Method
log10ws	-1.40		Crippen Method
logp	1.249		Crippen Method
mcvol	94.240	ml/mol	McGowan Method
pc	3633.35	kPa	Joback Method
ripol	1540.00		NIST Webbook
ripol	1540.00		NIST Webbook
tb	406.21	K	Joback Method
tc	595.13	K	Joback Method
tf	212.06	K	Joback Method
vc	0.358	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	170.12	J/molxK	406.21	Joback Method
cpg	178.73	J/molxK	437.70	Joback Method
cpg	187.00	J/molxK	469.18	Joback Method

cpg	194.91	J/mol×K	500.67	Joback Method
cpg	202.49	J/mol×K	532.16	Joback Method
cpg	209.74	J/mol×K	563.64	Joback Method
cpg	216.65	J/mol×K	595.13	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4245378&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4245378&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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