

# Ethane, 1,1,2,2-tetramethoxy-

<b>Other names:</b>	1,1,2,2-Tetramethoxyethane Glyoxal bis(dimethyl acetal) Tetramethoxy-ethane
<b>Inchi:</b>	InChI=1S/C6H14O4/c1-7-5(8-2)6(9-3)10-4/h5-6H,1-4H3
<b>InchiKey:</b>	IVXUXKRSTIMKOE-UHFFFAOYSA-N
<b>Formula:</b>	C6H14O4
<b>SMILES:</b>	COC(OC)C(OC)OC
<b>Mol. weight [g/mol]:</b>	150.17
<b>CAS:</b>	2517-44-4

## Physical Properties

Property code	Value	Unit	Source
gf	-425.24	kJ/mol	Joback Method
hf	-706.61	kJ/mol	Joback Method
hfus	9.00	kJ/mol	Joback Method
hvap	37.81	kJ/mol	Joback Method
log10ws	0.10		Crippen Method
logp	0.224		Crippen Method
mcvol	118.880	ml/mol	McGowan Method
pc	2934.52	kPa	Joback Method
tb	425.48	K	Joback Method
tc	598.88	K	Joback Method
tf	216.30	K	Joback Method
vc	0.431	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	249.82	J/mol×K	425.48	Joback Method
cpg	260.19	J/mol×K	454.38	Joback Method
cpg	270.41	J/mol×K	483.28	Joback Method
cpg	280.48	J/mol×K	512.18	Joback Method
cpg	290.36	J/mol×K	541.08	Joback Method
cpg	300.03	J/mol×K	569.98	Joback Method

cpg	309.48	J/mol×K	598.88	Joback Method
dvisc	0.0037667	Paxs	216.30	Joback Method
dvisc	0.0014955	Paxs	251.16	Joback Method
dvisc	0.0007437	Paxs	286.03	Joback Method
dvisc	0.0004305	Paxs	320.89	Joback Method
dvisc	0.0002774	Paxs	355.75	Joback Method
dvisc	0.0001933	Paxs	390.62	Joback Method
dvisc	0.0001429	Paxs	425.48	Joback Method
hvapt	42.90	kJ/mol	391.50	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2517444&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2517444&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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