

# Benzene, 1,3-dimethoxy-2-heptyl-5-methyl

<b>Inchi:</b>	InChI=1S/C16H26O2/c1-5-6-7-8-9-10-14-15(17-3)11-13(2)12-16(14)18-4/h11-12H,5-10H
<b>InchiKey:</b>	YYXVOYXNLRXXHG-UHFFFAOYSA-N
<b>Formula:</b>	C16H26O2
<b>SMILES:</b>	CCCCCCCc1c(OC)cc(C)cc1OC
<b>Mol. weight [g/mol]:</b>	250.38

## Physical Properties

Property code	Value	Unit	Source
gf	-42.64	kJ/mol	Joback Method
hf	-435.89	kJ/mol	Joback Method
hfus	32.45	kJ/mol	Joback Method
hvap	60.29	kJ/mol	Joback Method
log10ws	-5.08		Crippen Method
logp	4.525		Crippen Method
mcvol	224.280	ml/mol	McGowan Method
pc	1597.44	kPa	Joback Method
rinpol	1784.00		NIST Webbook
rinpol	1784.00		NIST Webbook
tb	651.94	K	Joback Method
tc	841.67	K	Joback Method
tf	378.52	K	Joback Method
vc	0.860	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	604.00	J/molxK	651.94	Joback Method
cpg	621.71	J/molxK	683.56	Joback Method
cpg	638.60	J/molxK	715.18	Joback Method
cpg	654.67	J/molxK	746.81	Joback Method
cpg	669.92	J/molxK	778.43	Joback Method
cpg	684.35	J/molxK	810.05	Joback Method
cpg	697.97	J/molxK	841.67	Joback Method
dvisc	0.0007852	Paxs	378.52	Joback Method

dvisc	0.0004526	Paxs	424.09	Joback Method
dvisc	0.0002903	Paxs	469.66	Joback Method
dvisc	0.0002014	Paxs	515.23	Joback Method
dvisc	0.0001483	Paxs	560.80	Joback Method
dvisc	0.0001143	Paxs	606.37	Joback Method
dvisc	0.0000914	Paxs	651.94	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R142848&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R142848&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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