

# Dipropyldiazene, trans

<b>Other names:</b>	(E)-1,2-Dipropyldiazene
<b>Inchi:</b>	InChI=1S/C6H14N2/c1-3-5-7-8-6-4-2/h3-6H2,1-2H3
<b>InchiKey:</b>	LQCYZZANGSLXDJ-UHFFFAOYSA-N
<b>Formula:</b>	C6H14N2
<b>SMILES:</b>	CCCN=NCCC
<b>Mol. weight [g/mol]:</b>	114.19
<b>CAS:</b>	55204-42-7

## Physical Properties

Property code	Value	Unit	Source
hf	-119.95	kJ/mol	Joback Method
hvap	35.62	kJ/mol	Joback Method
ie	8.10	eV	NIST Webbook
ie	8.61	eV	NIST Webbook
log10ws	-1.59		Crippen Method
logp	2.259		Crippen Method
mcvol	111.060	ml/mol	McGowan Method
pc	2365.67	kPa	Joback Method
rinpol	768.90		NIST Webbook
rinpol	768.90		NIST Webbook
tb	485.88	K	Joback Method
tc	684.90	K	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55204427&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55204427&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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