

# 1,5-Naphthalenedimethanol, alpha,alpha,alpha',alpha'-tetramethyl-

Inchi:	InChI=1S/C16H20O2/c1-15(2,17)13-9-5-8-12-11(13)7-6-10-14(12)16(3,4)18/h5-10,17-18
InchiKey:	QZIRPTFOJIXAPH-UHFFFAOYSA-N
Formula:	C16H20O2
SMILES:	CC(C)(O)c1cccc2c(C(C)(C)O)cccc12
Mol. weight [g/mol]:	244.33
CAS:	24168-56-7

## Physical Properties

Property code	Value	Unit	Source
gf	15.68	kJ/mol	Joback Method
hf	-290.87	kJ/mol	Joback Method
hfus	20.83	kJ/mol	Joback Method
hvap	87.22	kJ/mol	Joback Method
log10ws	-4.58		Crippen Method
logp	3.295		Crippen Method
mcvol	204.820	ml/mol	McGowan Method
pc	2535.37	kPa	Joback Method
tb	799.00	K	Joback Method
tc	1009.08	K	Joback Method
tf	480.72	K	Joback Method
vc	0.761	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	608.58	J/molxK	799.00	Joback Method
cpg	620.57	J/molxK	834.01	Joback Method
cpg	631.87	J/molxK	869.03	Joback Method
cpg	642.57	J/molxK	904.04	Joback Method
cpg	652.78	J/molxK	939.05	Joback Method
cpg	662.59	J/molxK	974.07	Joback Method
cpg	672.12	J/molxK	1009.08	Joback Method
dvisc	0.0005691	Paxs	480.72	Joback Method
dvisc	0.0001875	Paxs	533.77	Joback Method

dvisc	0.0000755	Paxs	586.81	Joback Method
dvisc	0.0000354	Paxs	639.86	Joback Method
dvisc	0.0000186	Paxs	692.91	Joback Method
dvisc	0.0000107	Paxs	745.95	Joback Method
dvisc	0.0000066	Paxs	799.00	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24168567&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24168567&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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