

1,2-dichloroheptane

Inchi:	InChI=1S/C7H14Cl2/c1-2-3-4-5-7(9)6-8/h7H,2-6H2,1H3
InchiKey:	WBCSJABQCRFUCP-UHFFFAOYSA-N
Formula:	C7H14Cl2
SMILES:	CCCCC(CI)CCI
Mol. weight [g/mol]:	169.09
CAS:	10575-87-8

Physical Properties

Property code	Value	Unit	Source
gf	-18.24	kJ/mol	Joback Method
hf	-224.57	kJ/mol	Joback Method
hfus	18.76	kJ/mol	Joback Method
hvap	53.20	kJ/mol	NIST Webbook
log10ws	-3.17		Crippen Method
logp	3.413		Crippen Method
mvol	133.970	ml/mol	McGowan Method
pc	2613.74	kPa	Joback Method
ripol	1366.00		NIST Webbook
ripol	1366.00		NIST Webbook
tb	433.98	K	Joback Method
tc	616.82	K	Joback Method
tf	213.49	K	Joback Method
vc	0.519	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	251.14	J/mol×K	433.98	Joback Method
cpg	262.79	J/mol×K	464.45	Joback Method
cpg	273.93	J/mol×K	494.93	Joback Method
cpg	284.57	J/mol×K	525.40	Joback Method
cpg	294.73	J/mol×K	555.87	Joback Method
cpg	304.42	J/mol×K	586.35	Joback Method
cpg	313.66	J/mol×K	616.82	Joback Method

dvisc	0.0074245	Paxs	213.49	Joback Method
dvisc	0.0029334	Paxs	250.24	Joback Method
dvisc	0.0014701	Paxs	286.99	Joback Method
dvisc	0.0008619	Paxs	323.74	Joback Method
dvisc	0.0005634	Paxs	360.48	Joback Method
dvisc	0.0003984	Paxs	397.23	Joback Method
dvisc	0.0002988	Paxs	433.98	Joback Method
hvapt	49.00	kJ/mol	409.50	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.44465e+01
Coeff. B	-3.95841e+03
Coeff. C	-6.93500e+01
Temperature range (K), min.	348.92
Temperature range (K), max.	502.67

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C10575878&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions

hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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