

# (±)-2-octyl hydrogen phthalate

<b>Other names:</b>	2-((Octan-2-yloxy)carbonyl)benzoic acid
<b>Inchi:</b>	InChI=1S/C16H22O4/c1-3-4-5-6-9-12(2)20-16(19)14-11-8-7-10-13(14)15(17)18/h7-8,10-
<b>InchiKey:</b>	DUJBEEUYBSLMGM-UHFFFAOYSA-N
<b>Formula:</b>	C16H22O4
<b>SMILES:</b>	CCCCCCC(C)OC(=O)c1cccc1C(=O)O
<b>Mol. weight [g/mol]:</b>	278.34
<b>CAS:</b>	68296-97-9

## Physical Properties

Property code	Value	Unit	Source
gf	-315.48	kJ/mol	Joback Method
hf	-663.40	kJ/mol	Joback Method
hfus	35.80	kJ/mol	Joback Method
hvap	86.34	kJ/mol	Joback Method
log10ws	-4.82		Crippen Method
logp	3.901		Crippen Method
mcvol	227.420	ml/mol	McGowan Method
pc	2012.70	kPa	Joback Method
rinpol	2143.00		NIST Webbook
rinpol	2143.00		NIST Webbook
tb	819.04	K	Joback Method
tc	1019.27	K	Joback Method
tf	476.93	K	Joback Method
vc	0.867	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	679.33	J/mol×K	819.04	Joback Method
cpg	735.19	J/mol×K	985.90	Joback Method
cpg	725.69	J/mol×K	952.52	Joback Method
cpg	715.38	J/mol×K	919.15	Joback Method
cpg	704.24	J/mol×K	885.78	Joback Method
cpg	692.23	J/mol×K	852.41	Joback Method

cpg	743.90	J/molxK	1019.27	Joback Method
dvisc	0.0000193	Paxs	819.04	Joback Method
dvisc	0.0000283	Paxs	762.02	Joback Method
dvisc	0.0000441	Paxs	705.00	Joback Method
dvisc	0.0000742	Paxs	647.99	Joback Method
dvisc	0.0001380	Paxs	590.97	Joback Method
dvisc	0.0002933	Paxs	533.95	Joback Method
dvisc	0.0007462	Paxs	476.93	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C68296979&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C68296979&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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