

# Methyl biphenyl-4-carboxylate

<b>Other names:</b>	Methyl-4-phenylbenzoate Methyl 4-biphenylcarboxylate p-Phenylbenzoic acid methyl ester [1,1'-Biphenyl]-4-carboxylic acid, methyl ester Methyl p-phenylbenzoate Methyl (1,1'-biphenyl)-4-carboxylate 4-Biphenylcarboxylic acid, methyl ester
<b>Inchi:</b>	InChI=1S/C14H12O2/c1-16-14(15)13-9-7-12(8-10-13)11-5-3-2-4-6-11/h2-10H,1H3
<b>InchiKey:</b>	GATUGNVDXMYTJX-UHFFFAOYSA-N
<b>Formula:</b>	C14H12O2
<b>SMILES:</b>	<chem>COC(=O)c1ccc(-c2ccccc2)cc1</chem>
<b>Mol. weight [g/mol]:</b>	212.24
<b>CAS:</b>	720-75-2

## Physical Properties

Property code	Value	Unit	Source
gf	48.27	kJ/mol	Joback Method
hf	-115.50	kJ/mol	Joback Method
hfus	22.50	kJ/mol	Joback Method
hvap	61.13	kJ/mol	Joback Method
log10ws	-4.32		Crippen Method
logp	3.140		Crippen Method
mcvol	168.040	ml/mol	McGowan Method
pc	2862.74	kPa	Joback Method
tb	654.35	K	Joback Method
tc	898.02	K	Joback Method
tf	390.00 ± 3.00	K	NIST Webbook
vc	0.627	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	415.01	J/mol×K	654.35	Joback Method
cpg	479.16	J/mol×K	857.41	Joback Method

cpg	468.50	J/molxK	816.80	Joback Method
cpg	456.81	J/molxK	776.19	Joback Method
cpg	444.02	J/molxK	735.57	Joback Method
cpg	430.11	J/molxK	694.96	Joback Method
cpg	488.82	J/molxK	898.02	Joback Method
dvisc	0.0001499	Paxs	654.35	Joback Method
dvisc	0.0001880	Paxs	609.47	Joback Method
dvisc	0.0002444	Paxs	564.59	Joback Method
dvisc	0.0003325	Paxs	519.71	Joback Method
dvisc	0.0004795	Paxs	474.82	Joback Method
dvisc	0.0007464	Paxs	429.94	Joback Method
dvisc	0.0012880	Paxs	385.06	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C720752&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C720752&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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